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(54) Title: SUBSTITUTED BETA-LACTAM COMPOUNDS USEFUL AS HYPOCHOLESTEROLEMIC AGENTS AND PROCESSES FOR THE PREPARATION THEREOF

Novel compounds of formula (I) wherein A is -CH = CH - B; -C = C - B;  $-(CH_2)_p - X - B$ , wherein p is 0-2 and X is a bond, -NH- or  $-S(O)_{0-2}$ ; optionally substituted heteroaryl or benzofused heteroaryl; -C(O) - L; or (Ia), wherein k is 1-2; D is B'- $(CH_2)_mC(O)$ -, wherein m is 1-5; B'- $(CH_2)_q$ -, wherein q is 2-6; B'- $(CH_2)_e$ -Z- $(CH_2)_r$ -, wherein Z is -O-, -C(O)-, phenylene, -NR<sub>8</sub>- or -S(O)<sub>0-2</sub>-, e is 0-5 and r is 1-5, provided that the sum of e and r is 1-6; B'-(alkenylene)-; B'-(alkadienylene)-; B'-(CH<sub>2)t</sub>-Z-(alkenylene), wherein t is 0-3, provided that the sum of t and the number of carbon atoms in the alkenylene chain is 2-6; B'-(CH<sub>2</sub>)<sub>1</sub>-V-(CH<sub>2</sub>)<sub>2</sub>-, wherein V is cycloalkylene, f is 1-5 and g is 0-5, provided that the sum of f and g is 1-6; B'-(CH<sub>2</sub>)<sub>t</sub>-V-(alkenylene) or B'-(alkenylene)-V-(CH<sub>2</sub>)<sub>t</sub>-, provided that the sum of t and the number of carbon atoms in the alkenylene chain is 2-6; B'-(CH<sub>2</sub>)<sub>a</sub>-Z-(CH<sub>2</sub>)<sub>b</sub>-V-(CH<sub>2</sub>)<sub>d</sub>-, wherein a, b and d are 0-6, provided that the sum of a, b and d is 0-6; T-(CH<sub>2</sub>)<sub>s</sub>-, wherein T is cycloalkyl and s is 1-6; naphthylmethyl or optionally substituted heteroarylmethyl; B is optionally substituted phenyl; B' is naphthyl, optionally substituted heteroaryl or optionally substituted phenyl; R is hydrogen, fluoro, alkyl, alkenyl, alkynyl, or B-(CH<sub>2</sub>)h-, wherein h is 0-3; R<sub>4</sub> is optionally substituted phenyl, indanyl, benzofuranyl, tetrahydronaphthyl, pyridyl, pyriazinyl, pyrimidinyl or quinolyl; are disclosed, as well as their use as hypocholesterolemic agents; the method of using compounds of formula (II), wherein R<sub>20</sub> is optionally substituted phenyl, optionally substituted naphthyl, optionally substituted heteroaryl, or optionally substituted benzofused heteroaryl, R21, R22 and R23 are independently selected from H or R<sub>20</sub>; E, F and G are independently a bond; cycloalkylene; alkylene; alkynylene; a substituted alkylene, alkenylene or alkynylene chain; an interrupted alkylene, alkenylene or alkynylene chain; or an interrupted alkylene, alkenylene or alkynylene chain substituted by one or more substituents; or one of R21-E and R22-F is selected from the group consisting of halogeno, OH, alkoxy, -OC(O)R<sub>5</sub>, -NR<sub>10</sub>R<sub>11</sub>, -SH or -S(alkyl); R<sub>5</sub> is alkyl, phenyl, R<sub>14</sub>-phenyl, benzyl or R<sub>14</sub>-benzyl; R<sub>10</sub> and R<sub>11</sub> are independently selected from H and lower alkyl, or a pharmaceutically acceptable salt thereof, in a pharmaceutically acceptable carrier as hypocholesterolemic agents is also disclosed.

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# SUBSTITUTED BETA-LACTAM COMPOUNDS USEFUL AS HYPOCHOLESTEROLEMIC AGENTS AND PROCESSES FOR THE PREPARATION THEREOF

## BACKGROUND OF THE INVENTION

The present invention relates to substituted  $\beta$ -lactams useful as hypocholesterolemic agents in the treatment and prevention of atherosclerosis and to processes for preparing  $\beta$ -lactams.

Atherosclerotic coronary heart disease represents the major cause for death and cardiovascular morbidity in the western world. Risk factors for atherosclerotic coronary heart disease include hypertension, diabetes mellitus, family history, male sex, cigarette smoking and serum cholesterol. A total cholesterol level in excess of 225-250 mg/dl is associated with significant elevation of risk.

Cholesteryl esters are a major component of atherosclerotic lesions and the major storage form of cholesterol in arterial wall cells. Formation of cholesteryl esters is also a key step in the intestinal absorption of dietary cholesterol. The intracellular esterification of cholesterol is catalyzed by the enzyme acyl CoA:cholesterol acyl transferase (ACAT, EC 2.3.1.26). Thus, inhibition of ACAT is likely to inhibit the progression of atherosclerotic lesion formation, decrease the accumulation of cholesteryl esters in the arterial wall, and block the intestinal absorption of dietary cholesterol.

A few β-lactam compounds have been reported as being useful in lowering cholesterol and/or in inhibiting the formation of cholesterol-containing lesions in mammalian arterial walls. U.S. 4,983,597 discloses N-sulfonyl-2-azetidinones as anticholesterolemic agents and Ram, et al., in <u>Indian J Chem. Sect. B. 29B</u>, 12 (1990), p. 1134-7, disclose ethyl 4-(2-oxoazetidin-4-yl)phenoxy-alkanoates as hypolipidemic agents. European Patent Publication 264,231 discloses

1-substituted-4-phenyl-3-(2-oxoalkylidene)-2-azetidinones as blood platelet aggregation inhibitors.

# SUMMARY OF THE INVENTION

Novel hypocholesterolemic compounds of the present invention are represented by the formula I

wherein

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A is -CH=CH-B;

-C≡C-B;

-(CH<sub>2</sub>) $_p$ -X-B, wherein p is 0, 1 or 2 and X is a bond, -NH-

or -S(O)<sub>0-2</sub>-;

heteroaryl, benzofused heteroaryl, W-substituted heteroaryl or W-substituted benzofused heteroaryl, wherein heteroaryl is selected from the group consisting of pyrrolyl, pyridinyl, pyrimidinyl, pyrazinyl, triazinyl, imidazolyl, thiazolyl, pyrazolyl, thienyl, oxazolyl and furanyl, and for nitrogen-containing heteroaryls, the N-oxides thereof, and wherein W is 1-3 substituents on the ring carbon atoms selected from the group consisting of lower alkyl, hydroxy lower alkyl, lower alkoxy, alkoxyalkyl, alkoxyalkoxy, alkoxycarbonylalkoxy, (lower alkoxy-imino)lower alkyl, lower alkanedioyl, lower alkyl lower alkanedioyl, allyloxy, -CF3, -OCF3, benzyl, R14-benzyl, benzyloxy, R14-benzyloxy, phenoxy, R14-phenoxy, dioxolanyl, NO2, -NR10R11, NR10R11(lower alkyl)-, NR10R11(lower alkoxy)-, OH, halogeno, -NHC(O)OR5, -NHC(O)R5, R6O2SNH-, (R6O2S)2N-, -S(O)2NH2, -S(O)0-2R10, tert-butyldimethylsilyloxymethyl,

-C(O)R<sub>12</sub> and  $^{-CH_2-N}$  , and wherein the substituents on the substituted heteroaryl ring nitrogen atoms, when present, are selected from the group consisting of lower alkyl, lower alkoxy, -C(O)OR<sub>5</sub>, -

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 $C(O)R_5$ , OH,  $NR_{10}R_{11}$ (lower alkyl)-,  $NR_{10}R_{11}$ (lower alkoxy)-,  $-S(O)_2NH_2$  and 2-(trimethylsilyl)ethoxymethyl;

-C(O)-B; or 
$$-(CH2)k-N = R13, wherein k is 1 or 2;$$

D is B'- $(CH_2)_mC(O)$ -, wherein m is 1, 2, 3, 4 or 5;

B'- $(CH_2)_{q}$ -, wherein q is 2, 3, 4, 5 or 6;

B'-(CH<sub>2</sub>)<sub>e</sub>-Z-(CH<sub>2</sub>)<sub>r</sub>, wherein Z is -O-, -C(O)-, phenylene, -NR<sub>8</sub>- or -S(O)<sub>0-2</sub>-, e is 0, 1, 2, 3, 4 or 5 and r is 1, 2, 3, 4 or 5, provided that the sum of e and r is 1, 2, 3, 4, 5 or 6;

B'-(C<sub>2</sub>-C<sub>6</sub> alkenylene)-; B'-(C<sub>4</sub>-C<sub>6</sub> alkadienylene)-;

 $B'-(CH_2)_t-Z-(C_2-C_6$  alkenylene)-, wherein Z is as defined above, and wherein t is 0, 1, 2 or 3, provided that the sum of t and the number of carbon atoms in the alkenylene chain is 2, 3, 4, 5 or 6;

B'- $(CH_2)_f$ -V- $(CH_2)_g$ -, wherein V is C<sub>3</sub>-C<sub>6</sub> cycloalkylene, f is 1, 2, 3, 4 or 5 and g is 0, 1, 2, 3, 4 or 5, provided that the sum of f and g is 1, 2, 3, 4, 5 or 6;

 $B'-(CH_2)_t-V-(C_2-C_6$  alkenylene)- or  $B'-(C_2-C_6$  alkenylene)-  $V-(CH_2)_t-$ , wherein V and t are as defined above, provided that the sum of t and the number of carbon atoms in the alkenylene chain is 2, 3, 4, 5 or 6:

B'- $(CH_2)_a$ -Z- $(CH_2)_b$ -V- $(CH_2)_d$ -, wherein Z and V are as defined above and a ,b and d are independently 0, 1, 2, 3, 4, 5 or 6, provided that the sum of a, b and d is 0, 1, 2, 3, 4, 5 or 6;

T-(CH<sub>2</sub>)<sub>s</sub>-, wherein T is cycloalkyl of 3-6 carbon atoms and

25 s is 1, 2, 3, 4, 5 or 6; or

naphthylmethyl, heteroarylmethyl, or W-substituted heteroarylmethyl, wherein heteroaryl and W are as defined above;

B is

$$R_1$$
  $R_2$   $R_3$ 

B' is naphthyl, heteroaryl or W-substituted heteroaryl, wherein heteroaryl is as defined above, or

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R is hydrogen, fluoro,  $C_1$ - $C_{15}$  alkyl,  $C_1$ - $C_{15}$  alkenyl,  $C_1$ - $C_{15}$  alkynyl, or B-(CH<sub>2</sub>)<sub>h</sub>-, wherein h is 0, 1, 2, or 3;

R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> are independently selected from the group consisting of H, lower alkyl, hydroxy lower alkyl, lower alkoxy, alkoxyalkyl, alkoxyalkoxy, alkoxycarbonylalkoxy, (lower alkoxyimino)-lower alkyl, lower alkanedioyl, lower alkyl lower alkanedioyl, allyloxy, -CF<sub>3</sub>, -OCF<sub>3</sub>, benzyl, R<sub>14</sub>-benzyl, benzyloxy, R<sub>14</sub>-benzyloxy, phenoxy, R<sub>14</sub>-phenoxy, dioxolanyl, NO<sub>2</sub>, -NR<sub>10</sub>R<sub>11</sub>, NR<sub>10</sub>R<sub>11</sub>(lower alkyl)-, NR<sub>10</sub>R<sub>11</sub>(lower alkoxy)-, OH, o-halogeno, m-halogeno, -NHC(O)OR<sub>5</sub>, -NHC(O)R<sub>5</sub>, R<sub>6</sub>O<sub>2</sub>SNH-, (R<sub>6</sub>O<sub>2</sub>S)<sub>2</sub>N-, -S(O)<sub>2</sub>NH<sub>2</sub>, -S(O)<sub>0</sub>-2R<sub>10</sub>, tert-

butyldimethylsilyloxymethyl, -C(O)R<sub>12</sub>,  $-CH_2-N$   $R_{13}$  and

 $\stackrel{\text{O}}{=}$   $\stackrel{\text{C}}{=}$   $\stackrel{\text{R}_{13}}{=}$ , or  $\text{R}_1$  is hydrogen and  $\text{R}_2$  and  $\text{R}_3$ , together with adjacent carbon atoms to which they are attached, form a dioxolanyl ring;

R<sub>1</sub>', R<sub>2</sub>' and R<sub>3</sub>' are independently selected from the group consisting of H, lower alkyl, hydroxy lower alkyl, lower alkoxy, alkoxyalkoxy, alkoxycarbonylalkoxy, (lower alkoxyimino)-lower alkyl, lower alkanedioyl, lower alkyl alkanedioyl, allyloxy, -CF<sub>3</sub>, -OCF<sub>3</sub>, benzyl, R<sub>14</sub>-benzyl, benzyloxy, R<sub>14</sub>-benzyloxy, phenoxy, R<sub>14</sub>-phenoxy, dioxolanyl, NO<sub>2</sub>, -NR<sub>10</sub>R<sub>11</sub>, NR<sub>10</sub>R<sub>11</sub>(lower alkyl)-, NR<sub>10</sub>R<sub>11</sub>(lower alkoxy)-, OH, halogeno, -NHC(O)OR<sub>5</sub>, -NHC(O)R<sub>5</sub>, R<sub>6</sub>O<sub>2</sub>SNH-, (R<sub>6</sub>O<sub>2</sub>S)<sub>2</sub>N-, -S(O)<sub>2</sub>NH<sub>2</sub>, -S(O)<sub>0-2</sub>R<sub>10</sub>, tert-butyldimethyl-

silyloxymethyl, -C(O)R<sub>12</sub>,  $R_{13}$  and  $R_{13}$ , or R<sub>1</sub>· is hydrogen and R<sub>2</sub>· and R<sub>3</sub>·, together with adjacent carbon atoms to which they are attached, form a dioxolanyl ring;

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R4 is , wherein n is 0, 1, 2 or 3, indanyl, benzofuranyl, benzodioxolyl, tetrahydronaphthyl, pyridyl, pyrazinyl, pyrimidinyl or quinolyl;

 $R_5$  is lower alkyl, phenyl,  $R_{14}$ -phenyl, benzyl or  $R_{14}$ -benzyl;  $R_6$  is OH, lower alkyl, phenyl, benzyl,  $R_{14}$ -phenyl or  $R_{14}$ -benzyl;  $R_7$  is lower alkyl, lower alkoxy, OH, halogeno, -NR<sub>10</sub>R<sub>11</sub>, -NHC(O)OR<sub>5</sub>, -NHC(O)R<sub>5</sub>, NO<sub>2</sub>, -CN, -N<sub>3</sub>, -SH, -S(O)<sub>0-2</sub>-(lower alkyl), -COOR<sub>9</sub>, -CONR<sub>10</sub>R<sub>11</sub>, -COR<sub>12</sub>, phenoxy, benzyloxy, -OCF<sub>3</sub>, or tert-butyldimethylsilyloxy, and where n is 2 or 3, the  $R_7$  groups can be the same or different;

R<sub>8</sub> is H, lower alkyl, phenyl lower alkyl, or -C(O)R<sub>9</sub>;
R<sub>9</sub> is H, lower alkyl, phenyl or phenyl lower alkyl;
R<sub>10</sub> and R<sub>11</sub> are independently selected from H and lower alkyl;

15 R<sub>12</sub> is H, OH, alkoxy, phenoxy, benzyloxy,
-NR<sub>10</sub>R<sub>11</sub>, lower alkyl, phenyl or R<sub>14</sub>-phenyl;

R<sub>13</sub> is -O-, -CH<sub>2</sub>-, -NH- or -N(lower alkyl)-; and

R<sub>14</sub> is 1-3 groups independently selected from the group consisting of lower alkyl, lower alkoxy, -COOH, NO<sub>2</sub>, -NR<sub>10</sub>R<sub>11</sub>, OH or halogeno;

or a pharmaceutically acceptable salt thereof.

Preferred are compounds of formula I wherein R is H.

Another group of preferred compounds of formula I is that wherein D is

B'-(CH<sub>2</sub>)<sub>q</sub>-, B'-(CH<sub>2</sub>)<sub>e</sub>-Z-(CH<sub>2</sub>)<sub>r</sub>-, B'-(C<sub>2</sub>-C<sub>6</sub> alkenylene)-,

or B'-(CH<sub>2</sub>)<sub>f</sub>-V-(CH<sub>2</sub>)<sub>g</sub>-, wherein B', Z, V, q, e, r, f, and g are as defined above. A third group of preferred compounds of formula I is that wherein R<sub>4</sub> is phenyl, R<sub>7</sub>-substituted phenyl or indanyl. Still another group of preferred compounds of formula I is that wherein A is -(CH<sub>2</sub>)<sub>p</sub>-X-B,

wherein X, B and p are as defined above.

Especially preferred are compounds of formula I wherein D is: B'-(CH<sub>2</sub>)<sub>q</sub>-, wherein B' is phenyl and q is 3 or 4; B'-(CH<sub>2</sub>)<sub>e</sub>-Z-(CH<sub>2</sub>)<sub>r</sub> ,

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wherein B' is p-fluorophenyl or p-methoxyphenyl, e is zero, Z is -O-, and r is 2; B'-( $C_2$ - $C_6$  alkenylene)- is 3-phenyl-1-propenyl; or B'-( $CH_2$ )<sub>f</sub>-V-( $CH_2$ )<sub>g</sub>-, wherein B' is phenyl, f is 1, V is cyclopropylene, and g is zero. Also especially preferred are compounds of formula I wherein A is -( $CH_2$ )<sub>p</sub>-X-B wherein p is zero and X is a bond. Preferably R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> are selected from H, OH, -NO<sub>2</sub>, lower alkoxy, alkoxyalkoxy, lower alkyl lower alkandioyl, m-halogeno, NR<sub>10</sub>R<sub>11</sub>(lower alkoxy)-, allyloxy, phenoxy, alkoxycarbonylalkoxy and -C(O)R<sub>12</sub>. Compounds wherein R<sub>1</sub> and R<sub>3</sub> are each H and R<sub>2</sub> is in the para-position are more preferred.

 $R_7$  is preferably selected from lower alkyl, lower alkoxy, halogeno, -OCF<sub>3</sub>, lower alkylthio, -NR<sub>10</sub>R<sub>11</sub>, -CN, OH, and -COR<sub>12</sub>. More preferred are compounds wherein n is 1 and  $R_7$  is in the paraposition.

Especially preferred compounds of formula I, wherein R is hydrogen, are shown in the following Table 1:

D	A	R <sub>4</sub>
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	C <sub>6</sub> H <sub>5</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-OH-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CI-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> CH <sub>2</sub> -O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH3CH2O-C6H4-
p-F-C <sub>6</sub> H <sub>4</sub> -O-(CH <sub>2</sub> ) <sub>2</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-F-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	m-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CF <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub> -CH=CH-	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-(CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> )-O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> S-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-(CH <sub>2</sub> =CH-CH <sub>2</sub> -O)-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-(C <sub>6</sub> H <sub>5</sub> -O)-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-(CH <sub>3</sub> CH <sub>2</sub> O <sub>2</sub> C)-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -

D	Α	R <sub>4</sub>
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	D
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-(CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> -O)-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>5</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-((CH <sub>3</sub> CH <sub>2</sub> ) <sub>2</sub> N)- C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-(CH <sub>3</sub> CH <sub>2</sub> O)-C <sub>6</sub> H <sub>4</sub> -	C <sub>6</sub> H <sub>5</sub> -
CH <sub>2</sub> CH <sub>-</sub> CH-	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-((CH <sub>3</sub> ) <sub>2</sub> CH-O)-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
p-F-C <sub>6</sub> H <sub>4</sub> -O-(CH <sub>2</sub> ) <sub>2</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	C <sub>6</sub> H <sub>5</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	N≡C-⟨¯⟩
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-(CH <sub>3</sub> O <sub>2</sub> C)-C <sub>6</sub> H <sub>4</sub> -	C <sub>6</sub> H <sub>5</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	F
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	H <sub>3</sub> CO OCH <sub>3</sub>	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-(H3CC(O))-C6H4-
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	→CH <sub>3</sub>	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-(CH <sub>3</sub> OCH <sub>2</sub> O)-C <sub>6</sub> H <sub>4</sub> -	C <sub>6</sub> H <sub>5</sub> -
p-CH3O-C6H4-O-(CH2)2-	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	C <sub>6</sub> H <sub>5</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-(CH <sub>3</sub> CH <sub>2</sub> OC(O)CH <sub>2</sub> O)- C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-(CH <sub>3</sub> OC(O)-(CH <sub>2</sub> ) <sub>2</sub> - C(O)-O)-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-((CH <sub>3</sub> ) <sub>2</sub> N-(CH <sub>2</sub> ) <sub>3</sub> -O)- C <sub>6</sub> H <sub>4</sub> -	p-CH₃O-C <sub>6</sub> H₄-
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	p-HO-C <sub>6</sub> H <sub>4</sub> -	p-HO-C <sub>6</sub> H <sub>4</sub> -
	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	C <sub>6</sub> H <sub>5</sub> -

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The first-listed compound in the above table having the (3R,4S) absolute stereochemistry is more preferred.

This invention also relates to the use of a novel compound of formula I of the present invention as a hypocholesterolemic agent in a mammal in need of such treatment.

In another aspect, the invention relates to a pharmaceutical composition comprising a novel  $\beta$ -lactam of formula I of the present invention in a pharmaceutically acceptable carrier.

In still another aspect, the present invention relates to a pharmaceutical composition comprising a cholesterol-lowering effective amount of a compound having the structural formula II

wherein

R<sub>20</sub> is phenyl, W-substituted phenyl, naphthyl, W-substituted naphthyl, benzodioxolyl, heteroaryl, W-substituted heteroaryl, benzofused heteroaryl and W-substituted benzofused heteroaryl, wherein heteroaryl is selected from the group consisting of pyrrolyl, pyridinyl, pyrimidinyl, pyrazinyl, triazinyl, imidazolyl, thiazolyl, pyrazolyl, thienyl, oxazolyl and furanyl, and for nitrogen-containing heteroaryls, the N-oxides thereof;

 $\ensuremath{\mathsf{R}}_{21},\,\ensuremath{\mathsf{R}}_{22}$  and  $\ensuremath{\mathsf{R}}_{23}$  are independently selected from H or

W is 1 to 3 substituents independently selected from the group consisting of lower alkyl, hydroxy lower alkyl, lower alkoxy, alkoxyalkyl, alkoxyalkoxy, alkoxycarbonylalkoxy, (lower alkoxyimino)-lower alkyl, lower alkanedioyl, lower alkyl lower alkanedioyl, allyloxy, -CF<sub>3</sub>, -OCF<sub>3</sub>, benzyl, R<sub>14</sub>-benzyl, benzyloxy, R<sub>14</sub>-benzyloxy, phenoxy, R<sub>14</sub>-phenoxy, dioxolanyl, NO<sub>2</sub>, -NR<sub>10</sub>R<sub>11</sub>, NR<sub>10</sub>R<sub>11</sub>(lower alkyl)-, NR<sub>10</sub>R<sub>11</sub>(lower alkoxy)-, OH, halogeno, -NHC(O)OR<sub>5</sub>, -NHC(O)R<sub>5</sub>,

R<sub>6</sub>O<sub>2</sub>SNH-, (R<sub>6</sub>O<sub>2</sub>S)<sub>2</sub>N-, -S(O)<sub>2</sub>NH<sub>2</sub>, -S(O)<sub>0-2</sub>R<sub>10</sub>, tert-butyldimethyl-

silyloxymethyl, -C(O)R<sub>12</sub>,  $-CH_2-N$   $R_{13}$  and -C-N  $R_{13}$ 

E, F and G are independently a bond; C3-C6 cycloalkylene; C1-C10 alkylene; C1-C10 alkenylene; C1-C10 alkynylene; an alkylene, alkenylene or alkynylene chain as defined substituted by one or more substituents independently selected from the group consisting of phenyl, W-substituted phenyl, heteroaryl and Wsubstituted heteroaryl, wherein heteroaryl is as defined above; an alkylene, alkenylene or alkynylene chain as defined interrupted by one or more groups independently selected from the group consisting of -O-. -S-, -SO-, -SO<sub>2</sub>-, -NR<sub>8</sub>, -C(O)-, C<sub>3</sub>-C<sub>6</sub> cycloalkylene, phenylene, Wsubstituted phenylene, heteroarylene and W-substituted heteroarylene: or an interrupted alkylene, alkenylene or alkynylene chain as defined substituted by one or more substituents independently selected from the group consisting of phenyl, W-substituted phenyl, heteroaryl and Wsubstituted heteroaryl; or one of R<sub>21</sub>-E and R<sub>22</sub>-F is selected from the group consisting of halogeno, OH, lower alkoxy, -OC(O)R5, -NR<sub>10</sub>R<sub>11</sub>, -SH or -S(lower alkyl);

 $R_5$  is lower alkyl, phenyl,  $R_{14}$ -phenyl, benzyl or  $R_{14}$ -benzyl;  $R_6$  is OH, lower alkyl, phenyl, benzyl,  $R_{14}$ -phenyl or  $R_{14}$ -

R<sub>8</sub> is H, lower alkyl, phenyl lower alkyl or -C(O)R<sub>9</sub>; R<sub>9</sub> is H, lower alkyl, phenyl or phenyl lower alkyl; R<sub>10</sub> and R<sub>11</sub> are independently selected from H and lower

25 alkyl;

benzyl;

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R<sub>12</sub> is H, OH, alkoxy, phenoxy, benzyloxy, -NR<sub>10</sub>R<sub>13</sub> -NR<sub>10</sub>R<sub>11</sub>, lower alkyl, phenyl or R<sub>14</sub>-phenyl;

R<sub>13</sub> is -O-, -CH<sub>2</sub>-, -NH- or -N(lower alkyl)-;

R<sub>14</sub> is 1-3 groups independently selected from the group consisting of lower alkyl, lower alkoxy, -COOH, NO<sub>2</sub>, -NR<sub>10</sub>R<sub>11</sub>, OH or halogeno;

provided that when G is a bond, R<sub>23</sub> is not H, and provided that when R<sub>23</sub> is W-substituted phenyl, W is not p-halogeno; or a pharmaceutically acceptable salt thereof; in a pharmaceutically acceptable carrier.

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It is noted that novel compounds of formula I are included within the scope of formula II.

Preferred compounds of formula II are those wherein  $R_{21}$  is H and E is a bond or lower alkylene, and those wherein  $R_{21}$  is phenyl and E is lower alkylene. Also preferred are compounds of formula II wherein  $R_{22}$  is H and F is a bond. Another group of preferred compounds of formula II is that wherein G is a bond and  $R_{23}$  is phenyl substituted by OH, -NO<sub>2</sub>, lower alkoxy, alkoxyalkoxy, m-halogeno, lower alkyl lower alkandioyl,  $NR_{10}R_{11}$  (lower alkoxy)-, allyloxy, phenoxy, alkoxycarbonylalkoxy, and -C(O) $R_{12}$ . Still another group of preferred compounds of formula II is that wherein  $R_{20}$  is phenyl or phenyl substituted by lower alkyl, lower alkoxy, halogeno, -OCF<sub>3</sub>, lower alkylthio, -NR<sub>10</sub>R<sub>11</sub>, -CN, OH or acetyl.

Certain compounds not within the scope of formula I but within the scope of formula II are novel compounds. Examples of such compounds are represented in the following Table 2

compounds are	represented in the	introversity rapic z	
R <sub>21</sub> -E-	R <sub>22</sub> -F-	R <sub>23</sub> -G-	R <sub>20</sub> -
C <sub>10</sub> H <sub>21</sub> -	Н	C <sub>6</sub> H <sub>5</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>10</sub> H <sub>21</sub> -	н	C <sub>6</sub> H <sub>5</sub> -	2,4,6-tri-CH <sub>3</sub> O- C <sub>6</sub> H <sub>2</sub> -
C <sub>10</sub> H <sub>21</sub> -	CH <sub>3</sub> CH <sub>2</sub> -	C <sub>6</sub> H <sub>5</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>10</sub> H <sub>21</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>10</sub> H <sub>21</sub> -	CH <sub>3</sub> CH <sub>2</sub> -	C <sub>6</sub> H <sub>5</sub> -	2,4,6-tri-CH <sub>3</sub> O- C <sub>6</sub> H <sub>2</sub> -
C <sub>10</sub> H <sub>21</sub> -	CH <sub>3</sub> CH <sub>2</sub> -	C <sub>6</sub> H <sub>5</sub> -CH=CH-	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
C <sub>10</sub> H <sub>21</sub> -	CH <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -

The present invention also relates to the method of lowering serum cholesterol in a mammal in need of such treatment

comprising administering a pharmaceutical composition comprising a compound of formula II in a pharmaceutically acceptable carrier.

The present invention also relates to the use of a compound of formula I or II as an ACAT inhibitor.

The present invention also relates to a stereoselective process for producing  $\beta$ -lactams of formula I, wherein R is hydrogen ,and D and A have trans relative stereochemistry, from a hydroxyamide of the formula

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wherein D, A and R<sup>4</sup> are as defined above, by cyclizing the hydroxyamide, and wherein the hydroxyamide is prepared from a carboxylic acid DCH<sub>2</sub>COOH, an aldehyde A-CHO and an amine R<sup>4</sup>NH<sub>2</sub>, wherein D, A and R<sup>4</sup> are as defined above, in a process utilizing as a chiral auxiliary an oxazolidinone of the formula

wherein R<sup>18</sup> and R<sup>19</sup> are independently selected from the group consisting of: hydrogen, C<sub>1</sub>-C<sub>6</sub> alkyl, phenyl, naphthyl, substituted phenyl, substituted naphthyl, and benzyl, wherein said chiral auxiliary is preferably R-(+)-4-benzyl-oxazolidinone.

This process, designated Method D, for producing compounds of formula I, wherein R is hydrogen, and D and A have trans relative stereochemistry, comprises the steps:

(a) reacting a carboxylic acid of the formula D-CH<sub>2</sub>COOH, wherein D is as defined above, with a chlorinating agent;

step (a);  (c) enolizing the product of step (b) with either:  (i) a dialkylboron triflate and a tertiary amine base; or  (ii) TiCl₄ and tetramethylethylenediamine (TMEDA) or a mixture of TMEDA and triethylamine;  then condensing with an aldehyde of the formula A-CHO, wherein A is as defined above;  (d) hydrolyzing the product of step (c) with a base and hydrogen peroxide;  (e) condensing the product of step (d) with an amine of the formula R⁴NH₂, wherein R⁴ is as defined above, by treating with a dehydrative coupling agent, optionally adding an activating agent;  (f) cyclizing the product of step (e) by reacting the product of step (e) with:  (i) a dialkylazodicarboxylate and a trialkylphosphine; or  (ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase transfer catalyst; or  (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention provides the steps of producing a β-lactam of formula I, wherein R is		tertiary amine base	deprotonating a chiral oxazolidinone, as described R-(+)-4-benzyloxazolidinone, with a strong base or a and treating the resulting anion with the product of
then condensing with an aldehyde of the formula A-CHO, wherein A is as defined above; (d) hydrolyzing the product of step (c) with a base and hydrogen peroxide; (e) condensing the product of step (d) with an amine of the formula R <sup>4</sup> NH <sub>2</sub> , wherein R <sup>4</sup> is as defined above, by treating with a dehydrative coupling agent, optionally adding an activating agent; (f) cyclizing the product of step (e) by reacting the product of step (e) with: (i) a dialkylazodicarboxylate and a trialkylphosphine; or (ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase 25 transfer catalyst; or (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or (iii) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention	5	• • •	(i) a dialkylboron triflate and a tertiary amine base;
wherein A is as defined above;  (d) hydrolyzing the product of step (c) with a base and hydrogen peroxide;  (e) condensing the product of step (d) with an amine of the formula R4NH2, wherein R4 is as defined above, by treating with a dehydrative coupling agent, optionally adding an activating agent;  (f) cyclizing the product of step (e) by reacting the product of step (e) with:  (i) a dialkylazodicarboxylate and a trialkylphosphine; or  (ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase transfer catalyst; or  (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention			(ii) TiCl <sub>4</sub> and tetramethylethylenediamine (TMEDA) or a mixture of TMEDA and triethylamine;
hydrogen peroxide;  (e) condensing the product of step (d) with an amine of the formula R <sup>4</sup> NH <sub>2</sub> , wherein R <sup>4</sup> is as defined above, by treating with a dehydrative coupling agent, optionally adding an activating agent;  (f) cyclizing the product of step (e) by reacting the product of step (e) with:  (i) a dialkylazodicarboxylate and a trialkylphosphine; or  (ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase and a phase transfer catalyst; or  (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention	10		
hydrogen peroxide;  (e) condensing the product of step (d) with an amine of the formula R <sup>4</sup> NH <sub>2</sub> , wherein R <sup>4</sup> is as defined above, by treating with a dehydrative coupling agent, optionally adding an activating agent;  (f) cyclizing the product of step (e) by reacting the product of step (e) with:  (i) a dialkylazodicarboxylate and a trialkylphosphine; or  (ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase and a phase transfer catalyst; or  (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention			
(e) condensing the product of step (d) with an amine of the formula R <sup>4</sup> NH <sub>2</sub> , wherein R <sup>4</sup> is as defined above, by treating with a dehydrative coupling agent, optionally adding an activating agent;  (f) cyclizing the product of step (e) by reacting the product of step (e) with:  (i) a dialkylazodicarboxylate and a trialkylphosphine; or  (ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase transfer catalyst; or  (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention		` '	
the formula R <sup>4</sup> NH <sub>2</sub> , wherein R <sup>4</sup> is as defined above, by treating with a dehydrative coupling agent, optionally adding an activating agent;  (f) cyclizing the product of step (e) by reacting the product of step (e) with:  (i) a dialkylazodicarboxylate and a trialkylphosphine; or  (ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase transfer catalyst; or  (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention		-	e;
dehydrative coupling agent, optionally adding an activating agent;  (f) cyclizing the product of step (e) by reacting the product of step (e) with:  (i) a dialkylazodicarboxylate and a trialkylphosphine; or  (ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase and a phase transfer catalyst; or  (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention			
(f) cyclizing the product of step (e) by reacting the product of step (e) with:  (i) a dialkylazodicarboxylate and a trialkylphosphine; or  (ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase transfer catalyst; or  (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention	15	the formula R4NH;	2, wherein R4 is as defined above, by treating with a
product of step (e) with:  (i) a dialkylazodicarboxylate and a  trialkylphosphine; or  (ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase transfer catalyst; or  (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention			ng agent, optionally adding an activating agent,
(i) a dialkylazodicarboxylate and a trialkylphosphine; or (ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase transfer catalyst; or (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention		• •	
trialkylphosphine; or  (ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase transfer catalyst; or  (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention		product of step (e)	
(ii) a di- or tri-chlorobenzoyl chloride, an aqueous solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase transfer catalyst; or  (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention			••
solution of a base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase transfer catalyst; or (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention	20		• • •
then treating the resulting di- or tri-chlorobenzoate with an aqueous solution of a base and a phase transfer catalyst; or (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention			
with an aqueous solution of a base and a phase transfer catalyst; or (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention			
transfer catalyst; or  (iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention			with an aqueous solution of a base and a phase
(iii) a dialkylchlorophosphate, an aqueous solution of a base and a phase transfer catalyst; or (iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention	<b>^</b> -		
of a base and a phase transfer catalyst; or  (iv) a di- or tri-chlorobenzoyl chloride and a metal  hydride.  In another embodiment, the process of this invention	25		
(iv) a di- or tri-chlorobenzoyl chloride and a metal hydride.  In another embodiment, the process of this invention			
hydride.  In another embodiment, the process of this invention			(iv) a di- or tri-chlorobenzoyl chloride and a metal
30 In another embodiment, the process of this invention			
provides the steps of producing a β-lactam of formula I, wherein R is	20	in ar	
	30	provides the steps	of producing a β-lactam of formula I, wherein R is
hydrogen, and D and A have trans relative stereochemistry, from a β-		hydrogen, and D a	and A have trans relative stereochemistry, from a β-
aminoamide derivative of the formula			

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$$\mathbb{R}^6$$
 $\mathbb{R}^7$ 
 $\mathbb{R}^7$ 
 $\mathbb{R}^8$ 
 $\mathbb{R}^8$ 
 $\mathbb{R}^8$ 
 $\mathbb{R}^8$ 
 $\mathbb{R}^8$ 

by cyclizing the β-aminoamide, wherein the β-aminoamide is prepared from a carboxylic acid DCH<sub>2</sub>COOH, and an imine ACH=N-R<sup>4</sup>, wherein D, A and R<sup>4</sup> are as defined above, in a process utilizing as a chiral auxiliary an oxazolidinone of the formula

wherein R<sup>18</sup> and R<sup>19</sup> are as defined above, and wherein said chiral auxiliary is preferably R-(+)-4-benzyl-oxazolidinone.

This process, designated Method F, for producing compounds of formula I, wherein R is hydrogen, and D and A have trans relative stereochemistry, comprises the steps:

- (a) enolizing the product of Method D, step (b) with TiCl<sub>4</sub> and tetramethylethylenediamine (TMEDA), then condensing with an imine of the formula A-CH=N-R<sup>4</sup>, wherein A and R<sup>4</sup> are as defined above:
- (b) cyclizing the product of step (a) by treating with astrong non-nucleophilic base, preferably sodium bistrimethylsilylamide.

#### **DETAILED DESCRIPTION:**

As used herein, the term "lower alkyl" means straight or branched alkyl chains of 1 to 6 carbon atoms and "lower alkyoxy" similarly refers to alkoxy groups having 1 to 6 carbon atoms.

"Alkenyl" means straight or branched carbon chains having one or more double bonds in the chain, conjugated or unconjugated,

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and alkadienyl refers to chains having two double bonds in the chain. Similarly, "alkynyl" means straight or branched carbon chains having one or more triple bonds in the chain.

Where an alkyl, alkenyl or alkynyl chain joins two other variables and is therefore bivalent, the terms alkylene, alkenylene and alkynylene are used.

"Cycloalkyl" means a saturated carbon ring of 3 to 6 carbon atoms, while "cycloalkylene" refers to a corresponding bivalent ring, wherein the points of attachment to other groups include all positional isomers.

"Halogeno" refers to fluorine, chlorine, bromine or iodine radicals.

"Heteroaryl" includes all positional isomers for a given heteroaryl group as defined above, for example 2-pyridyl, 3-pyridyl and 4-pyridyl. Benzofused heteroaryl refers to radicals formed by the bonding of a benzene radical to adjacent carbon atoms on a heteroaryl ring; examples are indolyl, quinolyl, quinazolinyl, quinoxalinyl, benzotriazolyl, indazolyl, benzoxazolyl, benzothienyl and benzofuranyl.

"Phenylene" means a bivalent phenyl group, including ortho, meta and para-substitution and "heteroarylene" similarly means a bivalent heteroaryl group, ioncluding all positional isomers.

"(Lower alkoxyimino)lower alkyl" refers to the group (C<sub>1</sub>-C<sub>6</sub> lower alkoxy)-N=CH-(C<sub>1</sub>-C<sub>5</sub> lower alkyl). "Lower alkanedioyl" means radicals of the formula -OC(O)(CH<sub>2</sub>)<sub>1-4</sub>C(O)OH, while "lower alkyl lower alkanedioyl" means radicals of the formula -OC(O)(CH<sub>2</sub>)<sub>1-4</sub>C(O)O-(lower alkyl).

 $R_{14}$ -benzyl and  $R_{14}$ -benzyloxy refer to benzyl and benzyloxy radicals which are substituted on the phenyl ring.

"Tertiary amine base" means a trialkylamine, such as triethylamine or diisopropylethylamine, or a nitrogen containing heterocycle, such as pyridine.

"Base" means a metal hydroxide base such as lithium, sodium or potassium hydroxide

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"Strong base" means a non-aqueous base such as a metal hydride or an alkyllithium.

"Metal hydride" means a commercially available metal hydride such as lithium, sodium or potassium hydride

"Alkyllithium" means a alkyllithium reagent such as n-butyllithium, s-butyllithium, t-butyllithium or methyllithium.

""Dehydrative coupling agent" means a carbodiimide such as 1-(3'-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (DEC) or dicyclohexylcarbodiimide (DCC).

"Activating agent" means an agent used to facilitate the formation of amide bonds such as 1-hydroxybenzotriazole (HOBT) or N-hydroxysuccinimide.

"Halide salt" means a metal salt of a halogen such as sodium, lithium or potassium bromide.

The carbon chains as defined in E, F, and G, when substituted by optionally substituted phenyl or heteroaryl groups, may include independent substitution on different carbon atoms, disubstitution on one carbon atom, or both. One skilled in the art will recognize that the number of double or triple bonds present, the replacement of carbon atoms in the chain and the presence of substitutents on the carbon atoms in the chain are all dependent on the length of the chain: shorter carbon chains cannot accommodate as many double or triple bonds, carbon replacements or substituents as longer carbon chains can. In general, unsaturated carbon chains contain 1 to 4 double or triple bonds, conjugated or non-conjugated. Where carbon atoms are replaced, 1 to 4 replacement groups can be present. Similarly, when carbon atoms in the chain are substituted, 1 to 4 substituents can be present.

Examples of alkyl chains are methyl, ethyl, propyl, butyl and decyl.

Examples of unsaturated E, F and G groups are ethylene and acetylene.

Examples of E, F and G groups wherein the carbon atoms in the chain are replaced are -CH<sub>2</sub>CH<sub>2</sub>O-, -OCH<sub>2</sub>CH<sub>2</sub>-, -CH<sub>2</sub>O-,

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-CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>O-, -CH<sub>2</sub>-O-CH<sub>2</sub>-, -CH<sub>2</sub>CH<sub>2</sub>-O-CH<sub>2</sub>, -CH<sub>2</sub>CH<sub>2</sub>-NH-, -CH<sub>2</sub>CH<sub>2</sub>-N(CH<sub>3</sub>)- and -O-CH<sub>2</sub>C(O)-NH-.

Compounds of the invention have at least one asymmetrical carbon atom and therefore all isomers, including diastereomers and rotational isomers are contemplated as being part of this invention. The invention includes d and I isomers in both pure form and in admixture, including racemic mixtures. Isomers can be prepared using conventional techniques, either by reacting enantiomeric starting materials or by separating isomers of a compound of formula I or II.

Isomers may include geometric isomers, e.g. when E, F or G contains a double bond. All such isomers are contemplated for this invention.

Those skilled in the art will appreciate that for some compounds of formulae I and II, one isomer will show greater pharmacological activity than another isomer.

Compounds of the invention with an amino group can form pharmaceutically acceptable salts with organic and inorganic acids. Examples of suitable acids for salt formation are hydrochloric, sulfuric, phosphoric, acetic, citric, oxalic, malonic, salicylic, malic, fumaric, succinic, ascorbic, maleic, methanesulfonic and other mineral and carboxylic acids well known to those in the art. The salt is prepared by contacting the free base form with a sufficient amount of the desired acid to produce a salt. The free base form may be regenerated by treating the salt with a suitable dilute aqueous base solution such as dilute aqueous sodium bicarbonate. The free base form differs from its respective salt form somewhat in certain physical properties, such as solubility in polar solvents, but the salt is otherwise equivalent to its respective free base forms for purposes of the invention.

Certain compounds of the invention are acidic (e.g., those compounds which possess a carboxyl group). These compounds form pharmaceutically acceptable salts with inorganic and organic bases. Examples of such salts are the sodium, potassium, calcium, aluminum, gold and silver salts. Also included are salts formed with

pharmaceutically acceptable amines such as ammonia, alkyl amines, hydroxyalkylamines, N-methylglucamine and the like.

Compounds of formulae I and II can be prepared by several known methods, and in particular compounds having trans stereochemistry can be prepared by a novel method D, as disclosed in U.S. Serial No. 07/734,426, filed July 23, 1991, or novel method F.

#### Method A:

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Compounds of formula la and lb, wherein A, D, R and R<sub>4</sub> are as defined above, can be prepared by treatment of an ester of formula V, wherein R<sub>15</sub> is lower alkyl such as ethyl or a chiral moiety such as menthyl or 10-(diisopropylsulfonamido)-isobornyl, with a strong base such as lithium diisopropylamide in a suitable solvent such as tetrahydrofuran (THF) at -78°C. Hexamethylphosphoric triamide (HMPA) may optionally be added as a cosolvent. An imine of formula IV is added and the reaction mixture is warmed to room temperature and the product isolated using conventional purification techniques. When a chiral ester of formula V is used, the resulting compound of formula la or lb is not racemic.

Compounds of formula Ic or Id can be converted to compounds of formula Ia or Ib by treatment with a strong base such as lithium diisopropylamide in a suitable solvent such as THF in the presence or absence of HMPA at -78°C, followed by the addition of an alkylating agent R-R<sub>17</sub>, or an acylating agent such as R-C(O)O-alkyl or R-C(O)Cl, wherein R is as defined above, except that R is not hydrogen, and R<sub>17</sub> is a leaving group such as bromo or iodo.

#### Method B':

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Trans compounds of formula Id can be converted to the corresponding cis compounds of formula Ic by using Method B, above, but using a proton source such as acetic acid in place of the alkylating agent.

#### 15 Method C:

Trans compounds of formula Id, wherein R is hydrogen and A, D and R<sub>4</sub> are as defined above can be prepared by treating cis compounds of formula Ic with a strong base such as lithium diisopropylamide or potassium t-butoxide in a suitable solvent such as THF. Those skilled in the art will appreciate that since the reaction conditions of Method C can be similar to those of Method A, the conversion of cis to trans sometimes occurs in situ under the conditions of Method A.

XXI

## Method D

Step D5

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Step D6

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In step D1 of Method D, the carboxylic acid XIII is treated with a chlorinating agent, e.g., thionyl chloride or oxalyl chloride, under a dry atmosphere, neat or in a suitable inert organic solvent, e.g., toluene, at 70°C to produce compound XIV.

In step D2, compound XV is converted to compound XVI in a two step reaction, first by deprotonating with a strong base, such as an alkyllithium, e.g., n-butyllithium, or metal hydride, e.g., sodium hydride, or a tertiary amine base, such triethylamine, in a suitable anhydrous organic solvent, e.g., dry THF, under a dry, inert atmosphere, e.g., nitrogen, at about 0°C to about -85°C, preferably about -78°C, over a period of about 30 to about 90 minutes, preferably about 60 minutes, and second by reacting the resulting anion, without isolation, with compound XIV in a suitable anhydrous organic solvent, e.g., dry THF, under a dry, inert atmosphere, e.g., nitrogen, at about -50°C to about -85°C, preferably about -78°C, over a period of about 30 to about 60 minutes, preferably about 45 minutes, followed by continued reaction at about -10°C to about 10°C, preferably about 0°C, for a period of about 30 to about 90 minutes, preferably about 60 minutes, then isolating the product, compound XVI, by extraction.

In step D3, compound XVI is treated with a dialkylboron triflate, e.g., di-n-butylborontriflate (Bu<sub>2</sub>BSO<sub>3</sub>CF<sub>3</sub>), in a suitable inert organic solvent, e.g. CH<sub>2</sub>Cl<sub>2</sub>, under a dry, inert atmosphere, e.g., nitrogen, at about -60°C to about 10°C, preferably about -10°C to about 0°C, for a period of about 10 minutes. A tertiary amine base, e.g., diisopropylethylamine, is added at about -10°C to about 0°C, preferably

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about -6°C to about-3°C, for about 20 to about 40 minutes, preferably about 30 minutes. The mixture is stirred at about -50°C to about -85°C, preferably about -78°C, for about 20 to about 40 minutes, preferably about 30 minutes, then treated with compound XVII, at about -50°C to about -85°C, preferably about -78°C, for about 20 to about 40 minutes, preferably about 30 minutes. The mixture is stirred at about -10°C to about 5°C, preferably about 0°C for about 30 to about 90 minutes, preferably about 60 minutes, then quenched with an aqueous pH 7 buffer solution, e.g., an aqueous solution of KH<sub>2</sub>PO<sub>4</sub> and NaOH, and treated with methanol, and hydrogen peroxide, preferably about 0°C, for about 1 hour. The product is isolated by extraction and crystallized from a suitable solvent, e.g., hexane/ethyl acetate, to obtain compound XVIII.

Alternatively, step D3 comprises treating compound XVI with titanium tetrachloride (TiCl<sub>4</sub>), in a suitable inert organic solvent, e.g. 15 CH<sub>2</sub>Cl<sub>2</sub>, at about -60°C to about 0°C, preferably about -25°C to about -15°C, and most preferably at about -20°C, for a period of about 10 minutes. Tetramethylethylenediamine (TMEDA) or a combination of TMEDA and triethylamine is added slowly over a period of about 10 minutes, while maintaining the temperature at about about -25°C to 20 about -10°C. The mixture is stirred at about -25°C to about -10°C, preferably about -15°C to about -10°C, for a period of 30 to 90 minutes, preferably about 60 minutes, then treated with a compound of the formula XVII. The mixture is stirred at about -25°C to about -10°C, preferably about -15°C to about -10°C, for a period of 30 to 90 minutes, 25 preferably about 60 minutes, then stirred for 30 to 60 minutes, preferably about 40 minutes, while warming to about 0°C to about 10°C, preferably about 10°C. The mixture is quenched with an aqueous solution of tartaric acid, preferably a solution of about 10% tartaric acid in water. The product is then isolated by extraction with a suitable solvent, e.g. 30 ethyl acetate, and recrystallized from a suitable solvent, such as ethyl actetate/hexane, to obtain compound XVIII.

In step D4, compound XVIII is treated with hydrogen peroxide, preferably 30% hydrogen peroxide, in a inert organic solvent,

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e.g., THF/ water, at about -5°C to about 5°C, preferably about 0°C, for about 10 to about 20 minutes, preferably about 15 minutes, then treated with a base, e.g., lithium hydroxide, at about -5°C to about 5°C, preferably about 0°C, until no starting material remains, as determined by thin layer chromatography (TLC), in about 2 hours to about 4 hours, preferably about 3 hours. The excess peracid is reduced by slowly adding a solution of sodium sulfite in water to the mixture over a period of about 30 to about 90 minutes, preferably about 70 minutes. The bulk of the solvent is removed under vacuum and the residue diluted with water. Compound IV is recovered from the mixture by extraction with a suitable inert organic solvent, e.g., toluene. The remaining aqueous solution is acidified to a pH of about 2.0 to about 3.0, preferably pH of about 2.4, using hydrochloric acid. The product is isolated by extraction using a suitable inert organic solvent, e.g., ethyl acetate, to provide compound XIX.

In step D5, compound XIX is reacted with compound XX, a dehydrative coupling agent, e.g., dicyclohexylcarbodiimide (DCC), and an activating agent, e.g., 1-hydroxybenzotriazole (HOBT), in a suitable inert organic solvent, e.g., dimethylformamide or acetonitrile, at about 25°C to about 50°C, preferably about 40°C. The reaction is continued until the starting material is consumed, as determined by TLC, in about 4 hours. The product is isolated by extraction to obtain compound XXI.

In step D6, compound XXI is cyclized by treating with triphenylphosphine or preferably a trialkylphosphine, e.g., tri-n-butyl-phosphine, and a dialkylazodicarboxylate, e.g., diethylazodicarboxylate (DEAD), in a suitable anhydrous organic solvent, e.g., dry THF, under a dry, inert atmosphere, e.g. nitrogen, at about -50°C to about -85°C, preferably about -70°C, for about 1 to about 3 hours, preferably about 2 hours. The reaction is then continued at about room temperature for about 12 to about 24 hours. The product is purified by preparative high performance liquid chromatography (HPLC) to obtain a compound of formula I, having trans relative stereochemistry. The use of tributylphosphine in this step gives reaction yields significantly higher than the yields obtained using triphenylphosphine.

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Alternatively, step D6 comprises combining compound XXI and a suitable phase transfer catalyst, e.g. tetra-n-butylammonium hydrogen sulfate, in a suitable solvent, such as methylene chloride. The mixture is stirred while cooling to about 0°C to 20°C, preferably about 10°C to about 20°C, then treated with an aqueous base, such as an alkali metal hydroxide, preferably 50% aqueous sodium hydroxide. A di- or tri-chlorobenzoyl chloride, preferably 2,6-dichlorobenzoyl chloride or 2,4,6-trichlorobenzoyl chloride, is slowly added over a period of 20 to 60 minutes, preferably about 30 minutes. The mixture is stirred at about 0°C to about 25°C, preferably about 15°C to about 20°C, for a period of 2 to 4 hours, preferably about 3 hours, then poured into cold water. The organic layer is separated and washed with water to neutral pH. The dior tri-chlorobenzoate product is isolated by crystallization from methylene chloride/heptane. The di- or tri-chlorobenzoate product is combined with a suitable phase transfer catalyst, e.g benzyltriethylammonium chloride, in a suitable solvent, such as a mixture methylene chloride and methyl t-butyl ether. The mixture is stirred at about 0°C to about 25°C, preferably about 15°C to about 20°C, and treated with an aqueous base, e.g. an alkali metal hydroxide, preferably 50% aqueous sodium hydroxide. After stirring for a period of 2 to 6 hours, preferably about 4 hours, the mixture is poured into ice water. The organic layer is washed with water to neutral pH. The product is isolated by removing the solvent, then purified by chromatography and recrystallization from a suitable solvent to give a compound of formula I, having trans relative stereochemistry.

A third alternative for step D6 comprises treating compound XXI in a suitable solvent, such as CH<sub>2</sub>Cl<sub>2</sub>, with a dialkylchlorophosphate, preferably diethylchlorophosphate, and an aqueous base, such as an alkali metal hydroxide, preferably 50% aqueous sodium hydroxide, in the presence of phase transfer catalyst, such as tetra-n-butylammonium hydrogen sulfate or benzyltriethylammonium chloride.

Another alternative for step D6 comprises treating compound XXI with a di- or tri-chlorobenzoyl chloride, preferably 2,6-

dichlorobenzoyl chloride or 2,4,6-trichlorobenzoyl chloride, and a suitable base, such as sodium hydride in a suitable solvent, such as CH<sub>2</sub>Cl<sub>2</sub>, dimethylformamide, or a combination thereof. The product is isolated and purified by chromatography followed by crystallization from a suitable solvent, e.g. ether/hexane.

Starting compounds XIII, XV, XVII, and XX are all either commercially available or well known in the art and can be prepared via known methods.

#### Method E:

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Compounds of formula la can also be prepared by treatment of an imine of formula IV with an activated carboxylic acid derivative of formula XII in the presence of a base such as triethylamine, tributylamine or diethylisopropylamine in an inert solvent such as CH<sub>2</sub>Cl<sub>2</sub>, heptane or toluene. Examples of activated carboxylic acid derivatives of formula XII include acid chlorides (L=Cl), mixed anhydrides formed with phenyl phosphorodichloridate (L=OP(O)(Cl)OPh), and N-methylpyridinium esters formed from the reaction of an acid with N-methyl-2-chloropyridinium iodide (L=2-oxy-N-methylpyridinium iodide).

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Starting materials of formulae IV and V are known or can be prepared by methods well known in the art.

#### Method F:

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In step F1, compound XVI (from Method D, step 2) is dissolved in a suitable solvent, e.g. methylene chloride, then treated with titanium tetrachloride at about -60°C to about 0°C, preferably about -25°C to about -15°C, under a dry, inert atmosphere, preferably nitrogen, for a period of about 5 min. TMEDA is added and the mixture stirred at about -60°C to about -10°C, preferably about -25°C to about -20°C, for a period of about 1 hour. The imine (A-CH=N-R4) is slowly over a period of 20 to 40 min., preferably about 30 min., and the mixture is stirred at about -60°C to about 0°C, preferably about -25°C to about -15°C, for 20 to 40 min., preferably about 30 min. The mixture is then warmed to about 0°C and the reaction monitored by high pressure liquid chromatography (HPLC) until complete. The mixture is then poured into a solution of tartaric acid in water, preferably 10% tartaric acid. The product is isolated by extraction with a suitable solvent, e.g. ethyl acetate, then purified by crystallization.

In step F2, the product of step F1 is treated with a strong non-nucleophilic base, such as sodium or lithium bistrimethylsilylamide, in a suitable inert organic solvent, e.g. CH<sub>2</sub>Cl<sub>2</sub>, at about -20°C to about 10°C, preferably about 0°C. The mixture is stirred while gradually warming to about 20° to about 25°C, then monitored by HPLC until the starting material is gone, typically after a period of 1 to 2 hours. The reaction mixture is poured into aqueous tartaric acid, preferably 10% tartaric acid, and the product isolated from the organic layer.

Imines of the formula A-CH=N-R<sup>4</sup> can be prepared from aldehydes of the formula A-CHO and amines of the formula R<sup>4</sup>-NH<sub>2</sub> by procedures well known in the art. Aldehydes of formula A-CHO and amines of formula R<sup>4</sup>-NH<sub>2</sub> are commercially available or can be prepared via known procedures.

Compounds of formula II can be prepared by methods similar to those described for compounds of formula I.

It will also be apparent to those skilled in the art, and by reference to the examples which follow, that compounds of formulae I and II can be converted into different compounds of formula I or II by well known methods. For example, a compound of formula I wherein A comprises a double or triple bond, or a compound of formula II wherein G comprises a double or triple bond can be converted to the corresponding saturated compound by treatment with hydrogen gas in the presence of a catalyst such as palladium or platinum on carbon.

Reactive groups not involved in the above processes can be protected during the reactions with conventional protecting groups which can be removed by standard procedures after the reaction. The following Table 3 shows some typical protecting groups:

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Group to be Protected	Table 3 Group to be Protected and Protecting Group
-COOH	-COOalkyl, -COObenzyl, -COOphenyl
NH	NCOalkyl, NCObenzyl, NCOphenyl,
	NCH <sub>2</sub> OCH <sub>2</sub> CH <sub>2</sub> Si(CH <sub>3</sub> ) <sub>3</sub> , NC(O)OC(CH <sub>3</sub> ) <sub>3</sub> ,
	N-benzyl, NSi(CH <sub>3</sub> ) <sub>3</sub> , NSi-C(CH) <sub>3</sub>
	O CH3
⊸NH <sub>2</sub>	-N
	Ü CH₃
—ОН	$-OCH_3$ , $-OSi(CH_3)_3$ , $-OSi-C(CH)_3$ $CH_3$

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We have found that the compounds of this invention lower serum lipid levels, in particular serum cholesterol levels. Compounds of this invention have been found to inhibit the intestinal absorbtion of cholesterol and to significantly reduce the formation of liver cholesteryl esters in animal models. Some compounds also inhibit ACAT <u>in vitro</u>. Thus, compounds of this invention are hypocholesterolemic agents by virtue of their ability to inhibit the esterification and/or intestinal absorption of cholesterol; they are therefore useful in the treatment and prevention of atherosclerosis in mammals, in particular in humans.

In addition to the compound aspect, the present invention therefore also relates to a method of lowering serum cholesterol levels, which method comprises administering to a mammal in need of such treatment a hypocholesterolemic effective amount of a compound of formula I or II of this invention. The compound is preferably administered in a pharmaceutically acceptable carrier suitable for oral administration.

The <u>in vitro</u> and <u>in vivo</u> activity of the compounds of formulae I or II can be determined by the following procedures.

ACAT Assav (in vitro)

This assay measures the activity of ACAT by measuring the ACAT-mediated transfer of tritiated oleic acid from acyl-CoA to cholesterol to give labelled cholesteryl oleate. Rat liver microsomes are used as the source of ACAT. Assays are performed in round bottom microtiterplates using a total incubation volume of 50  $\mu L$ . Each incubation well receives 10  $\mu L$  assay buffer (0.5M KHPO4, 10  $\mu M$  dithiothreitol, pH 7.4), 7.5  $\mu L$  of 40 mg/mL BSA (Bovine Serum Albumin) and 12.5  $\mu g$  of microsomal protein. The test compound (in sufficient amount to bring the final concentration to from 0.1 to 25  $\mu M$ ), reference compound, or vehicle control is added and the final volume brought to 47  $\mu L$ . The microtiterplate is then floated on the surface of a 37°C water bath for fifteen minutes. Incubations are started by the addition of 3  $\mu L$  3H-acyl CoA (1 $\mu Ci/well$ , final concentration of 10  $\mu M$  acyl CoA). The plate is then returned to the water bath for 15 minutes. The incubations are then terminated by application of 15  $\mu L$  from each incubation to

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Standards are applied to several lanes so that the cholesteryl ester band can be identified. After drying, the plates are eluted with 90:10:1 petroleum ether:diethyl ether:acetic acid. The standards are visualized via iodine vapor, and the regions corresponding to cholesteryl ester are scraped into 7 mL scintillation vials. 4 mL of scintillant are added to each vial, and the radioactivity quantified. Background count is determined by the boiled controls. Full activity is determined by activity in the presence of vehicle. The percent inhibition is calculated by subtracting the background from both control and test samples, and the test value is calculated as a percentage of the control. For IC50 determinations, the inhibition is plotted against drug does on a log scale and the concentration at which 50% inhibition is obtained is determined.

## 15 <u>In Vivo Assay of Hypolipidemic Agents Using the Hyperlipidemic</u> Hamster

Hamsters are separated into groups of six and given a controlled cholesterol diet (Purina Chow #5001 containing 0.5% cholesterol) for seven days. Diet consumption is monitored to determine dietary cholesterol exposure in the face of test compounds. The animals are dosed with the test compound once daily beginning with the initiation of diet. Dosing is by oral gavage of 0.2mL of corn oil alone (control group) or solution (or suspension) of test compound in corn oil. All animals moribund or in poor physical condition are euthanized. After seven days, the animals are anesthetized by IM injection of ketamine and sacrificed by decapitation. Blood is collected into vacutainer tubes containing EDTA for plasma lipid analysis and the liver excised for tissue lipid analysis. Data is reported as percent reduction of lipid versus control.

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The present invention also relates to a pharmaceutical composition comprising a compound of formula I or II of this invention and a pharmaceutically acceptable carrier. The compounds of formula I or II can be administered in any conventional oral dosage form such as

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capsules, tablets, powders, cachets, suspensions or solutions. The formulations and pharmaceutical compositions can be prepared using conventional pharmaceutically acceptable excipients and additives and conventional techniques. Such pharmaceutically acceptable excipients and additives include non-toxic compatible fillers, binders, disintegrants, buffers, preservatives, anti-oxidants, lubricants, flavorings, thickeners, coloring agents, emulsifiers and the like.

The daily hypocholesteremic dose of a compound of formula I or II is about 7 to about 30 mg/kg of body weight per day. For an average body weight of 70kg, the dosage level is therefore from about 500 to about 2000 mg of drug per day, given in a single dose or 2-4 divided doses. The exact dose, however, is determined by the attending clinician and is dependent on the potency of the compound administered, the age, weight, condition and response of the patient.

Following are examples of preparing compounds of formulae I and II. The stereochemistry listed is relative stereochemistry unless otherwise noted. The terms cis/trans refer to the relative orientations at the beta-lactam 3- and 4- positions when each is monosubstituted (i.e., R=H).

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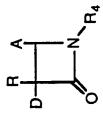
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Freshly prepare lithium diisopropyl amide (LDA) by dissolving 23.96mL (17.39g, 172mmol) diisopropylamine in 230mL dry THF at -78°C under nitrogen. Add 103.9mL (166 mmol @ 1.6M in hexanes) of *n*-butyl lithium and stir at -78°C for 1h. To this cold solution add 32.58g (158mmol) of 5-phenyl valeric acid ethyl ester in 195mL dry THF over ~1h, keeping the reaction temperature below -65°C. Stir for 1h at -78°C, then add 38.13g (158mmol) of 4-methoxybenzylidine

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anisidine in 350mL dry  $CH_2Cl_2$ . Allow the reaction to slowly come to room temperature and the precipitate that forms will begin to go into solution. Stir the reaction mixture for 16h at room temperature. Partition the mixture between 1.2liter of 1N aqueous hydrochloric acid (HCl) and 1liter of ether. Wash the ether layer with 300 mL 1N HCl. Combine the acid layers and extract with 1 liter of ether. Combine the ether extracts, dry over MgSO<sub>4</sub> and concentrate in vacuo. Crystallize the residue (35.08g, 55%) from ~200mL ethyl acetate-hexane (1:1) to give 32.05g of the title racemic compound as an off-white crystal, mp = 90-93°C.

Using a similar procedure, the following compounds shown in Tables 4 and 4A can also be prepared:



¥	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	H	C <sub>6</sub> H <sub>5</sub> -CH=CH-	38,48	p-MeO-C <sub>6</sub> H <sub>4</sub> -	mp=119.5-120.5°C
	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	p-OH-C <sub>6</sub> H <sub>4</sub> -	38,48	p-MeO-C <sub>6</sub> H <sub>4</sub> -	mp=152.5-155.0°C
₹	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	н	p-MeO-C <sub>6</sub> H <sub>4</sub> -	38,48	p-MeO-C <sub>6</sub> H <sub>4</sub> -	mp=86-88°C
			CH=CH-			
1P	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	3-MeO-C <sub>6</sub> H <sub>4</sub> -	38,48	p-MeO-C <sub>6</sub> H <sub>4</sub> -	mp=90.5-91.0°C
5	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	ェ	p-MeO-C <sub>6</sub> H <sub>4</sub> -	trans	3-pyridyl-	1H NMR (CDC13) 8 8.41
						(s,1H), 8.28(s, 1H), 7.79(d,
						1H), 7.20(m, 8H), 6.90(d,
						2H), 4.63(s, 1H), 3.80(s, 3H),
						3.20(t, 1H), 1.85(m, 4H)
1AN	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	ェ	p-MeO-C <sub>6</sub> H <sub>4</sub> -	trans	p-CF <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CF3O-C6H4- elemental analysis: calc'd
						C=68.56, H=5.31, N=3.08;
						found C=68.32, H=5.12,
						N=2.97
8	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-MeO-C <sub>6</sub> H <sub>4</sub> -	cis	p-CI-C <sub>6</sub> H <sub>4</sub> -	elemental analysis: calc'd
	, ,					C=73.97, H=5.96, N=3.45,
						Cl=8.73; found C=73.63,
						H=5.92, N=3.52, Cl=9.12

1AP	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-MeO-C <sub>6</sub> H <sub>4</sub> -	cis	p-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> -	elemental analysis: calc'd
						C=81.01, H=7.06, N=3.63;
						found C=80.97, H=7.06,
						N=3.74
1AQ	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	cis	p-C <sub>6</sub> H <sub>5</sub> O-	elemental analysis: calc'd
					C <sub>6</sub> H <sub>4</sub> -	C=80.32, H=6.31, N=3.02;
						found C=80.26, H=6.27,
						N=3.18
1AR	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	н	N	trans	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=101.5-102.5°C
1AS	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	(CH <sub>3</sub> ) <sub>3</sub> CSi(CH <sub>3</sub> ) <sub>2</sub> -	trans	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - EIMS (M+1)=516
			OCH2-C <sub>6</sub> H <sub>4</sub> -			
1AT	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	(CH <sub>3</sub> ) <sub>3</sub> CSi(CH <sub>3</sub> ) <sub>2</sub> -	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=84-85°C	mp=84-85°C
			OCH <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> -			
1AU	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	H	4-CH <sub>3</sub> S-C <sub>6</sub> H <sub>4</sub> -	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H;-	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>2</sub> - mp=103-104°C
1AV	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	P 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	trans	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - FABMS (M+1)=492
			CH <sub>2</sub> O(CH <sub>2</sub> ) <sub>2</sub> Si(CH <sub>3</sub> ) <sub>3</sub>			
1AW	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	PZ	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=108.5-110°C
			CH2O(CH2)2Si(CH3)3			

1AX	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I			p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - EIMS (M)=443	EIMS (M)=443
1AY	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-((CH <sub>3</sub> ) <sub>2</sub> CH-O)- C <sub>6</sub> H <sub>4</sub> -	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=96-97°C
1AZ	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-(CH <sub>3</sub> (CH) <sub>2</sub> -O)- C <sub>6</sub> H <sub>4</sub> -	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=96-97°C
1BA	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	HOO CH3	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=91.5-92.5°C
188	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> )3-	I	-0cH <sub>3</sub>	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=86.0-87.0°C
1BC	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> )3-	I	C	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=130.0-131.0°C
180	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	ェ	C	trans	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=111.0-112.5°C
1BE	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	Ç,	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=74.5-75°C
1BF	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	ェ	Ç	trans	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=79.5-80.5°C
186	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	H	4-CF <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> -	trans	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=94.5-96.0°C
18H	_	I	4-CF3-C <sub>6</sub> H <sub>4</sub> -	cls	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=114.0-115.5°C

181	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I		cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=117-120°C
			Y.S.			
187	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> )3-	エ	S	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -   mp=88.5-90.5°C
107	C.H. (CHo)	I	3-CF <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=66.0-68.0°C
20 0		I	3-CF <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - CI MS (M+1)=456.2
10L		I	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> -	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=124-125°C	mp=124-125°C
1 BN		I	4-(C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub> -O)-	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=131-132°C	mp=131-132°C
	_		C <sub>6</sub> H <sub>4</sub> -			
Cat	CoHz.(CHa)a-	I	CH3CH2-O-C6H4-	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=84-85°C
징윤		王	Ş	trans	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=68-70°C
			?	0000	n-CH2O-C2H4- mn=50-52°C	mn=50-52°C
18Q	C6H5-(CH2)3-	I_		2		
		]	CH2(CH2)2-O-CeH4-	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=71.0-72.5°C
된	+		4-(CeHe-O)-CeH4-	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=108-109°C
1BS	<del>-</del> +	= :	4 (O H-) C-H-	oic Sign	n-CH2O-CeH4-	mp=127.5-128.5°C
1BT	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> )3-	I	4-(0605)-0604-	25	11 0 0 110	
1BU	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-(CH2=CH-CH2-O)-	si Si	p-CH3O-C6H4-	0-6:11-0:11=dm
			C <sub>6</sub> H <sub>4</sub> -			

18V	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	Ť	cls	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=116-117°C	mp=116-117°C
			OCH <sub>3</sub>			
1BW	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	н	CI NCH <sub>3</sub>	cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=129.5-130.5°C
18X	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> )3-	н		cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=170.0-170.5°C
1BY	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	Н	4-(CH3CH2-O)-C6H4-	cis	C <sub>6</sub> H <sub>5</sub> -	mp=89.5-90.0°C
1BZ	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	н		cis	p-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=100-104°C
1CA	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I		trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=91-92°C	mp=91-92°C
1CB	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	H	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	cis	(CH3)3CSI(CH3)2- O-C6H4-	mp=93-94°C
1CC	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>2</sub> -	Н	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	cis	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=128-129°C
1CD	C <sub>6</sub> H <sub>5</sub> -O-(CH <sub>2</sub> ) <sub>2</sub> -	H	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	cis	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=120-121°C
1CE	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>6</sub> -	Ξ	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	cis	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=80-81°C	mp=80-81°C
1CF	CH <sub>2</sub> )3-	I	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	cis	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=101-102°C

10G	4-F-C6H4-O-(CH2)2-	I	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	cis	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH3O-C6H4- elemental analysis: calc'd
						C=71.25, H=5.74, N=3.32;
						found C=71.34, H=5.73,
						N=3.43
1CH	4-F-C6H4-O-(CH2)2-	I	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	cis	C <sub>6</sub> H <sub>5</sub> -	elemental analysis: calc'd
				, <del></del>		C=73.64, H=5.67, N=3.;
						found C=73.39, H=5.51,
						N=3.54
10	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>4</sub> -	I	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	cis	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH3O-C6H4- elemental analysis: calc'd
						C=78.04, H=7.03, N=3.37;
						found C=77.69, H=6.98,
						N=3.60
1CM	1CM C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	H	4-CH <sub>3</sub> S-C <sub>6</sub> H <sub>4</sub> -	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=104-105.5°C
1CN	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	Ê	cis	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=144.0-144.5 <sup>o</sup> C
55	1CO C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	3)2N-(CH2)3-O)-	cis	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=92-94°C	mp=92-94°C
			C <sub>6</sub> H <sub>4</sub> -			

-					-n-C	Project reference Leading
CeHs	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I_	4-CH3O-C6H4-	CIS CIS	-605-	elemental analysis: caic d
						C=80.83, H=6.78, N=3.77;
						found C=80.69, H=6.78,
						N=3.90
			В%			
				E V		
		R <sub>21</sub> −E−				
			-z'	F		
					Rel. Stereo-	
	B.4.F.	Boo-F-	R23-G-	R <sub>20</sub>	chemistry	Data
Hor	121 Hoa-	- 75. H	CeHs-	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3H,4R	mp=176.0-177.0°C
	C40H94-	I	C <sub>6</sub> H <sub>5</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3S,4R	mp=83.5-84.0°C
	C10H21-	Ξ	C <sub>6</sub> H <sub>5</sub> -	2,4,6-tri-MeO-	3R,4R	
<u> </u>		·		C <sub>6</sub> H <sub>4</sub> -	3.6:1 cls/trans	
]		S.H.	CeHs-	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3R,4S	mp=158.5-159.0°C
= =		1-Pr	CeHs-	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3R,4R	mp=162-164°C
ב כ		C.He.	CeHs-	p-MeO-C <sub>6</sub> H <sub>4</sub> -	38,48	mp=185.0-186.0°C
5 7	15-01-12-1	2.6F.	CeHs-	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3R,4S	mp=123.0-124.5°C
<u>ه</u>	C6H5-CH2-	100113	2:02			

12	CH <sub>3</sub> CH <sub>2</sub> .	C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> -	38,48	mp=101.0-102.5°C
1AA	CH <sub>3</sub> CH <sub>2</sub> .	C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3R,4S	mp=73.5-75.0°C
1AB	I	CH <sub>3</sub> CH <sub>2</sub> .	C <sub>6</sub> H <sub>5</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3R,4R	mp=134.0-135.5°C
1AC	cyc-C <sub>5</sub> H <sub>11</sub>	C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> -	8:5 mix cis/frans	mp=57.5-62.5°C
					phenyl	
1AD	I	CH3CH2-	C <sub>6</sub> H <sub>5</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3R,4S	mp=107-109°C
1AE	I	CH <sub>3</sub> CH <sub>2</sub> .	C <sub>6</sub> H <sub>5</sub> -	2,4,6-tri-MeO-	3H,4H	CI MS: (M+1)342
				C <sub>6</sub> H <sub>4</sub> -		
1AF	I	CH <sub>3</sub> CH <sub>2</sub> .	C <sub>6</sub> H <sub>5</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3H,4S	FAB MS:(M+1)307.9
			CH=CH-			
1AG	I	CH <sub>3</sub> CH <sub>2</sub> .	C <sub>6</sub> H <sub>5</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3S,4S	mp=84-86°C
			сн=сн-			
1AH	I	CH3-	C <sub>6</sub> H <sub>5</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3S,4S	mp=122-123°C
1AI	Н	CH <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3S,4R	mp=93-94°C
<b>1AJ</b>	I	снзсн2-	p-N02-	p-MeO-C <sub>6</sub> H <sub>4</sub> -	3S,4R	mp=115-116 <sup>0</sup> C
			C <sub>6</sub> H <sub>4</sub> -			
1AK	H	СН3СН2-	4-NO2-	4-MeO-C6H4-	3R,4R	mp=101-102 <sup>o</sup> C
			C <sub>6</sub> H <sub>4</sub> -			
1AL	H	снзснз-	4-MeO-	4-MeO-C <sub>6</sub> H <sub>4</sub> -	38,48	mp=96-103 <sup>o</sup> C
			C <sub>6</sub> H <sub>4</sub> -			

102	1CJ (CH3)2-N-	I	4-MeO- C6H4-	C <sub>6</sub> H <sub>5</sub> -	cis	mp=115-1170C
1CK	1CK C <sub>6</sub> H <sub>5</sub> -	снзсн2-	4-MeO- CeHa-	4-MeO-C6H4-	trans	mp=56-58 <sup>o</sup> C
1CL	1CL C10H21-	I	4-MeO- C6H4-	4-меО- СеН4-	cis	elem. anal.: calc'd C=76.56, H=8.80, N=3.31; found C=76.60, H=8.77, N=3.48

STEP 1: Add 39.1mL (117.2 mmol) of ethyl magnesium bromide to a 0°C solution of 31.00g (97.6 mmol) of (-)-10-(diisopropylsulfonamido)-isoborneol (See Oppolzer, et al, <u>Tet. Lett.. 25</u> (1984), p. 5885) in 370mL dry THF. Stir the mixture for 0.5h at 0°C, then at room temperature for 0.5h. Add 37.74mL (39.67g, 117.2mmol) of 5-phenylvaleric acid anhydride to this mixture at 0°C and stir overnight at room temperature. Pour the mixture into 1 liter of half-saturated NaHCO<sub>3</sub> and extract with two-800mL portions of hexane. Dry the combined hexane layers with Na<sub>2</sub>SO<sub>4</sub> and concentrate in vacuo. Chromatograph the residue (57.35g) in three portions over ~800g SiO<sub>2</sub> eluting with 2% ethyl acetate-CH<sub>2</sub>Cl<sub>2</sub> to obtain the desired ester (42.07g, 90.2%), FAB MS: (M+1) 479.

STEP 2: Using a procedure similar to Example 1, treat the ester of Step 1, washing the ether layer with 200mL 1N aqueous HCl, drying with MgSO<sub>4</sub> and concentrating in vacuo. Chromatograph the residue over 1kg SiO<sub>2</sub>, eluting with 20% ethyl acetate-hexane to give 24.19g (79%) of the recovered alcohol and 25.66 (66%) of the title enantiomerically enriched β-lactam. This lactam can be further enriched by chromatography over a Chiralcel<sup>TM</sup> OD column (Daicel Chemical Industries, Ltd., Fort Lee, NJ), eluting with 10% isopropanol-hexanes. Crystallize the resulting enantiomerically pure (3S,4S) compound from ether-hexane to obtain a white solid, mp = 84 - 85°C, [α]<sub>D</sub><sup>25</sup> = -98.0°

Using a similar procedure, the following enantiomerically pure (3R,4R) compound (EXAMPLE 2A) can also be prepared:

(m.p. = 84-85°C; FAB MS: (M+1) 402.2;  $[\alpha]_D^{25} = +98.0^\circ$  (MeOH))

#### **EXAMPLE 3**

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Freshly prepare a solution of lithium isopropylcyclohexyl amide (LICA) by dissolving 0.68mL (0.58g, 4.11mmol) isopropylcyclohexyl amine in 20mL THF at -78°C under nitrogen. Add 2.58mL (4.03mmol) *n*-butyl lithium (1.6M, from Aldrich, Milwaukee, WI) and stir at -78°C for 1h. Add 1.04g (2.61mmol) of the compound of Example 1K in 5mL dry THF to the solution. After 2h at -78°C, add 2.8mL (2.89g, 16mmol) of hexamethylphosphoric triamide followed by 0.33mL (641mg, 4.11mmol) of ethyl iodide at -78°C. Stir the mixture overnight at room temperature. Quench the reaction with 40mL 1N aqueous HCl and extract with two-50mL portions of CH<sub>2</sub>Cl<sub>2</sub>. Combine the CH<sub>2</sub>Cl<sub>2</sub> layers and wash sequentially with 50mL 1N aqueous HCl and 50mL Na<sub>2</sub>SO<sub>3</sub>. Dry over MgSO<sub>4</sub> and concentrate in vacuo. Chromatograph the residue over 40g SiO<sub>2</sub>, eluting with 20% ethyl acetate-hexane to give 0.95g (83%) of the title compound as a colorless oil, FAB MS: (M+1) 426.4 Using a similar procedure, the following compounds

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shown in Tables 5 and 5Acan also be prepared:

			R O	<b>∢</b> ——		
				N R4		
	-			Relative		
Ж	۵	<b>E</b>	۷	Stereo-	R4	Data
				chemistry		
34	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	CH3CH2-	C <sub>6</sub> H <sub>5</sub> -	38,48	p-MeO-C <sub>6</sub> H <sub>4</sub> -	mp=103.0-103.5°C
3B	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -	3R,4S	p-MeO-C <sub>6</sub> H <sub>4</sub> -	FAB MS: (M+1) 448
30	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>10</sub> H <sub>21</sub> -	C <sub>6</sub> H <sub>5</sub> -	3S,4R	p-MeO-C <sub>6</sub> H <sub>4</sub> -	CI MS: (M+1) 512
30	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	CH <sub>3</sub> CH <sub>2</sub> -	C <sub>6</sub> H <sub>5</sub> -	38,48	2,4,6-tri-MeO-	mp=145.5-147.0°C
					C <sub>6</sub> H₄-	
3E	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	CH <sub>3</sub> CH <sub>2</sub> - C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -	3R,4S	p-MeO-C <sub>6</sub> H₄-	mp=82.5-83.5°C
3F	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	CH <sub>3</sub> CH <sub>2</sub> -	C <sub>6</sub> H <sub>5</sub> -	3H,4S	2,4,6-tri-MeO-	mp=132-134°C
					C <sub>6</sub> H <sub>4</sub> -	
ဗ္ဗ	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	CH <sub>3</sub> CH <sub>2</sub> -	CH <sub>3</sub> CH <sub>2</sub> - C <sub>6</sub> H <sub>5</sub> -CH=CH-	3S,4S	p-MeO-C <sub>6</sub> H <sub>4</sub> -	mp=98-99°C
ЭН	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>2</sub> -	CH <sub>3</sub> CH <sub>2</sub> - C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -	3H,4S	p-MeO-C <sub>6</sub> H <sub>4</sub> -	mp=100.0-100.5°C
31	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>2</sub> -	СН3-	C <sub>6</sub> H <sub>5</sub> -	3R,4S	p-MeO-C <sub>6</sub> H <sub>4</sub> -	mp=130-131°C

נ	c	۵	٧	Relative	78	Data
<u>.</u>	נ	=		chemistry	<b>b</b> 1	
3)	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	СН3-	C <sub>6</sub> H <sub>5</sub> -	38,48	p-MeO-C <sub>6</sub> H <sub>4</sub> - mp=88-89°C	mp=88-89°C
3K	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>4</sub> -	CH3-	C <sub>6</sub> H <sub>5</sub> -	38,48	p-MeO-C <sub>6</sub> H <sub>4</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> - mp=109-110°C
31	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>4</sub> -	CH3CH2- C6H5-	C <sub>6</sub> H <sub>5</sub> -	38,48	p-MeO-C <sub>6</sub> H <sub>4</sub> -	20°5€-96°C
ЭМ	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	CH <sub>3</sub> CH <sub>2</sub> -	CH <sub>3</sub> CH <sub>2</sub> - p-MeO-C <sub>6</sub> H <sub>4</sub> -	3R,4S	p-MeO-C <sub>6</sub> H <sub>4</sub> -	FAB MS: (M+1) 430.4
3N	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>4</sub> -	C <sub>6</sub> H <sub>5</sub> -CH=CH-	racemic	p-MeO-C <sub>6</sub> H <sub>4</sub> -	FAB MS: (M+1) 516.4
		(CH <sub>2</sub> )3-				
30	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	CH3CH2-	CH <sub>3</sub> CH <sub>2</sub> - p-MeO-C <sub>6</sub> H <sub>4</sub> -	38,48	p-MeO-C <sub>6</sub> H <sub>4</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> - FAB MS: (M+1) 430.4

		Data	mp=47.5-48.5°C	mp=103.0-103.5°C	mp=48.0-49.5°C	CI MS: (M+1) 512.6		CI MS: (M+1) 482	2₀86-96=dш	EI MS: (M+) 425	mp=96-97°C
		R <sub>20</sub>	p-MeO-C <sub>6</sub> H <sub>4</sub> -		2,4,6-tri-MeO- C <sub>6</sub> H <sub>4</sub> -	2,4,6-tri-MeO- C <sub>6</sub> H <sub>4</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> -	p-MeO-C <sub>6</sub> H <sub>4</sub> -			
N. N	Rel. Stereo-	chemistry	38,48	3R,4S	3R,4R	3R,4R		3S,4R	3R.4R	38,48	38,48
R <sub>21</sub> - E - F - O		R <sub>23</sub> -G-	C <sub>6</sub> H <sub>5</sub> -		C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -CH=CH-	C <sub>6</sub> H <sub>5</sub> -			
		R <sub>22</sub> -F-	СН3СН2-	C <sub>6</sub> H <sub>5</sub> -	12-	C <sub>6</sub> H <sub>5</sub> -	(CH <sub>2</sub> ) <sub>3</sub> -	CH <sub>3</sub> CH <sub>2</sub> - C <sub>6</sub> H <sub>5</sub> -	CH <sub>3</sub> CH <sub>2</sub> - C <sub>6</sub> H <sub>5</sub> -	СН3СН2-	CH <sub>3</sub> -
		R21-E-	C <sub>10</sub> H <sub>21</sub> -	C <sub>10</sub> H <sub>21</sub> -	C <sub>10</sub> H <sub>21</sub> -	C10H21-		C <sub>10</sub> H <sub>21</sub> -	C <sub>10</sub> H <sub>21</sub> -	C10H21-	C10H21-
		Ex.	30	3R	3Т	30		30	же	3X	3¥

## **EXAMPLE 4**

Add 100mg (0.30mmol) of the trans  $\beta$ -lactam of Example 1V in 2mL dry THF to a solution of 1.22mL (0.32mmol) LDA (1.5M, from Aldrich, Milwaukee, WI) in 0.5mL dry THF at -78°C under nitrogen. Stir for 5min at -78°C, then quench at low temperature with ~0.3mL acetic acid. Partition the mixture between 30mL ethyl acetate and 20mL water. Wash the organic layer with 20 mL 10% aqueous NaHCO3 solution, dry over MgSO4 and concentrate in vacuo to give 98mg of an oily white solid. Chromatograph the residue over SiO<sub>2</sub>, eluting with 20% ethyl acetate-hexane to obtain 26 mg of the title cis  $\beta$ -lactam as a white solid, mp = 141.3 - 142.3°C.

# EXAMPLE 5

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Dissolve 32.05g (79.8mmol) of the racemic cis  $\beta$ -lactam of Example 1 in 500mL of THF. Add 1.79g (16.0mmol) of potassium tbutoxide and stir at 0°C for 1.5h. Partition the reaction mixture between 600mL 1N aqueous HCl and 1.2 liters of ether. Extract the aqueous layer with 400mL of ether. Combine the ether layers, dry over MgSO4 and concentrate in vacuo to give 32.0g of a mixture of cis and trans  $\beta\text{-}$ lactams (2.7:1). Isolate the pure trans  $\beta$ -lactam via silica gel HPLC

chromatography, eluting with 10% ethyl acetate-hexane. Crystallize from ethyl acetate-hexane to give white crystals, mp = 96.0 - 97.5°.

Using a procedure similar to that of Example 5, the following compounds shown in Tables 6 and 6A can be prepared:

	Data	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - FAB MS (M+1)=402.2	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=109.5-110.0°C	CIMS (M+1)=390	3-CH3O-C <sub>6</sub> H4- elemental analysis: calc'd	C=77.78, H=6.78, N=3.49; found	C=77.75, H=6.70, N=3.60	CIMS (M+1)=406	elemental analysis: calc'd	C=78.04, H=7.04, N=3.37; found	C=78.01, H=7.03, N=3.47
	R4	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-F-C <sub>6</sub> H <sub>4</sub> -	3-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -			4-CI-C <sub>6</sub> H <sub>4</sub> -	4-(CH <sub>3</sub> CH <sub>2</sub> -0)-	C6H4-	
	Relative Stereo- chemistry	3S,4R single	enantiomer 3R,4S	trans	trans			trans	trans		
O N N N N N N N N N N N N N N N N N N N	<b>∀</b>	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	3-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -			4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -		
	Н	I	I	I	I			H	I		
	D	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>9</sub> ) <sub>3</sub> -		C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -			C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -		
	Ex.	5B	50	5D	<b>SE</b>			5G	5H		

51	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	H	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	4-CH <sub>3</sub> S-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> S-C <sub>6</sub> H <sub>4</sub> - elemental analysis: calc'd
						C=74.79, H=6.52, N=3.35; found
						C=74.74, H=6.44, N=3.48
5.3	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	Н	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> -	elemental analysis: calc'd
						C=81.01, H=7.06, N=3.63; found
						C=81.25, H=7.09, N=3.75
5K	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	Н	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	4-(C <sub>6</sub> H <sub>5</sub> -O)-	elemental analysis: calc'd
				,	C <sub>6</sub> H <sub>4</sub> -	C=80.32, H=6.31, N=3.02; found
						C=80.36, H=6.29, N=3.16
5L	(СН2)3111	н	EHDO-{()}	trans	C <sub>6</sub> H <sub>5</sub> -	mp=74-75°C
	)		)	optically		$ \alpha _{D}^{25} = -7.6^{\circ}$
				pure		
5M	( ) (CH <sub>2</sub> )3	H	()	trans	C <sub>6</sub> H <sub>5</sub> -	mp=74-75°C
			)	optically		$\left  \left[ \alpha \right]_{D}^{25} = -7.5^{\circ} \right $
				pure		
5P	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	-{_}- ocH₃ ocH₃	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=70.5-71.5°C
20	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I		trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=79.5-81.0°C
			L			

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H.	C6H5-(CH2)3-	I	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> -	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -   mp=74.5-77.0°C
58	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-(CH <sub>3</sub> CH <sub>2</sub> -O)-	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=85-87°C	mp=85-87°C
			C <sub>6</sub> H <sub>4</sub> -			
5T	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	H	4-(C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub> -O)-	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=111-112°C
			C <sub>6</sub> H <sub>4</sub> -			•
20	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-(CH2=CH-CH2-0)-	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=75.5-77.0°C
			C <sub>6</sub> H <sub>4</sub> -			
50	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	H	4-(C6H5-O)-C6H4-	trans	4-CH3O-C <sub>6</sub> H <sub>4</sub> -	mp=103.5-105.5°C
5W	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>2</sub> -	I	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - m <sub>2</sub> =75-77°C	mp=75-77°C
2X	C <sub>6</sub> H <sub>5</sub> -O-(CH <sub>2</sub> ) <sub>2</sub> -	I	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - m <sub>0</sub> =122-123°C
5∀	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>6</sub> -	I	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH3O-C6H4- elemental analysis: calc'd
						C=78.52, H=7.50, N=3.18; found
[						C=78.32, H=7.37, N=3.25
70	C6H5-(CH2)2-O-	E	4-CH3O-C6H4-	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	elemental analysis; calc'd
						C=74.42, H=6.25, N=3.47; found
						C=74.57, H=6.29, N=3.51
5AA	4-F-C6H4-O-(CH2)2-	I_	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	4-MeO-C <sub>6</sub> H <sub>4</sub> -	elemental analysis: caic'd
						C=71.25, H=5.74, N=3.32; found
						C=71.34, H=5.73, N=3.43

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elemental analysis: calc'd C=78.04, H=7.03, N=3.37; found	46.0=1.30, U=0.31, U=0		Data	mp=113-115°C	mp=122-123°C	
4-MeO-C <sub>6</sub> H <sub>4</sub> -			R <sub>20</sub>	4-MeO-C <sub>6</sub> H <sub>4</sub> -	C <sub>6</sub> H <sub>5</sub> -	
trans		Rel. Stereo-	chemistry	3S,4R	trans	
4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	R <sub>21</sub> - E F		R <sub>23</sub> -G-	C <sub>6</sub> H <sub>5</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	
Ι	·		R <sub>22</sub> -F-	エ	I	
5AB C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>4</sub> -		1	R21-E-	(СН3)2СН-	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>2</sub> -	N(CH <sub>3</sub> )-
5AB			Ä.	5A	SN SN	

Dissolve 200mg (0.47mmol) of the β-lactam of Example 3G in 4mL EtOAc and add ~10mg 10% Pd/C. Hydrogenate at 1atm for 4h.

Filter the mixture through Celite and concentrate. Chromatograph the residue over 40g SiO<sub>2</sub>, eluting with 20% EtOAc-hexane to obtain 188mg (94%) of the title β-lactam as a colorless oil, CI MS: (M+1) 428.1.

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To a refluxing solution of 4.33 grams (.0205 moles) N-(4-methoxybenzylidine)-aniline and 7.6 grams (0.0410 moles) tributylamine in 40 mL heptane add,in portions, a solution of 4.03 grams (0.0205 moles) 5-phenylvaleroyl chloride in 15 mL heptane over about 2 hours, then reflux the solution for an additional 4 hours. Evaporate the solvent and take up the residue in 150 mL EtOAc. Wash with 1N HCl (2 x 30 mL), saturated NaHCO<sub>3</sub> (1 x 30 mL) and brine (1 x 30 mL), then dry over MgSO<sub>4</sub> and evaporate to give 7.69 grams of a semisolid. Recrystallization from 15% EtOAc in hexane to obtain 3.08 grams of the title compound, mp 75-76°C. An additional 2.31 g is obtained from the mother liquors by chromatography (silica gel, 5% ethyl acetate in hexane) and recrystallization.

Using a procedure similar to that of Example 7, the following compounds shown in Tables 7 and 7Acan be prepared:

•						
	Data	mp=75-77°C	elemental analysis: calc'd C=72.10, H=5.81, N=6.73; found C=72.46, H=5.91,	N=6.54 mp=91-93°C	mp=79-81°C	elemental analysis: calc'd C=73.69, H=5.69, N=3.44; found C=73.36, H=5.77, N=3.48
	R4	4-(CH <sub>3</sub> CH <sub>2</sub> OC(O))- C <sub>6</sub> H <sub>4</sub> -	3-NO <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> -	N=6.54 4-((CH <sub>3</sub> CH <sub>2</sub> ) <sub>2</sub> -N)- mp=91-93°C C <sub>6</sub> H <sub>4</sub> -	4-NC-C <sub>6</sub> H <sub>4</sub> -	
	Relative Stereo- chemistry	trans	trans	trans	trans	trans
4 N N N N N N N N N N N N N N N N N N N	V	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
	. Ж	Ξ	I	Ξ	Η	I
	Q	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -
	Ä.	7A	78	70	7D	7E

7F	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	エ	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	- OCH3	mp=103-104°C
76	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	4-(CH <sub>3</sub> C(O))-	elemental analysis: calo'd
					C <sub>6</sub> H <sub>4</sub> -	C=78.42, H=6.58, N=3.39;
						found C=78.08, H=6.51,
						N=3.40
7H	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	<u>"</u>	mp=90-92°C
					L OCH3	
<b>F</b>	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	CH <sub>3</sub>	mp=129-130°C
					L CH3	
₹	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-((CH <sub>3</sub> ) <sub>2</sub> CH-O)- C <sub>6</sub> H <sub>4</sub> -	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=96.5-97.5°C
7	ייחטי יחיט	-				
2	С6п5-(СП2 <i>)</i> 3-	<b>.</b>	L COCH3	trans	C <sub>6</sub> H <sub>5</sub> -	mp=89-91ºC
7P	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	Н	4-(CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> -O)-	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=92,5-93,5°C
			C <sub>6</sub> H <sub>4</sub> -		•	
۵۲ ا	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	-CH20 -CH3	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	EIMS (M)=415
			j g			
TH.	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	I	4-(CH3OC(O))-C <sub>6</sub> H <sub>4</sub> -	trans	C <sub>6</sub> H <sub>5</sub> -	mb=84-85°C
						2 22 3

	L					
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> - H	I		hoo 🔷	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - EIMS (M)=420
		4	ш			
工		4	4-(CH <sub>3</sub> OCH <sub>2</sub> O)-C <sub>6</sub> H <sub>4</sub> -	trans	C <sub>6</sub> H <sub>5</sub> -	CIMS (M+1)=402
Ξ		4	4-((CH <sub>3</sub> ) <sub>2</sub> N)-C <sub>6</sub> H <sub>4</sub> -	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	1
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> - H			— с — с — осн <sub>з</sub>	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	
С <sub>6</sub> Н <sub>5</sub> -(СН <sub>2</sub> ) <sub>3</sub> - H			- C- C- C- OCH <sub>3</sub>	cis	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - EIMS (M)=429
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> - H H <sub>3</sub>		f	H <sub>3</sub> CO	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - EIMS (M)=431	EIMS (M)=431
C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> - H 4-(C		4-(C	4-(CH <sub>3</sub> CH <sub>2</sub> -O)-C <sub>6</sub> H <sub>4</sub> -	trans	CeHe-	mp-05 0 05 500
4-O <sub>2</sub> N-C <sub>6</sub> H <sub>4</sub> -(CH <sub>2</sub> ) <sub>3</sub> - H 4-		4-	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>2</sub> O-C <sub>E</sub> H <sub>4</sub> - mn-136 E 137 000
С <sub>6</sub> Н <sub>5</sub> -(СН <sub>2</sub> )3- н 4-		4	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	trans	4-I-CeH4-	mn=96 F 07 FOC
С <sub>6</sub> Н <sub>5</sub> -(СН <sub>2</sub> )3- Н 4-		4-	4·(CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> -0)- CeH <sub>4</sub> -	trans	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	mp=79-82°C
CeHe-(CHa)a-O-	$\vdash$		1 0 0 10			
	$\frac{1}{2}$	4	4-CH3O-C6H4-	cis	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH3O-CeH4-   mp=149-1500C

				,		,	<del>,</del>					<del></del>				
elemental analysis; caic'd	C=78.17, H=6.31, N=3.51;	found C=78.22, H=6.36,	N=3.59	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - FAB mass spec	400(100)	mp=105-106°C	elemental analysis: calc'd	C=78.29, H=7.27, N=3.26;	found C=78.37, H=7.26,	N=3.41	mp=54-56°C	mp=55-57°C	elemental analysis: calc'd	C=74.42, H=6.25, N=3.47;	found C=74.13, H=6.03,	N=3.26
4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -				4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -		4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -				C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -			
trans				trans		trans	trans				trans	trans	trans			
4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -				4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -		4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -				4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -			
I				I		H	I				I	Н	I			
					,	(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>5</sub> -				4-0 <sub>2</sub> N-C <sub>6</sub> H4-0- (CH <sub>2</sub> ) <sub>2</sub> -		4-(CH <sub>3</sub> O)-C <sub>6</sub> H <sub>4</sub> -	O-(CH2)2-		
7AI				7A.		7AK	7AM				7AN	7A0	7AP			

									·			·	
<sup>1</sup> H NMR (CDCl <sub>3</sub> ) 7.97-7.78	(m,2H); 7.69 (d, J=8Hz,1H);	7.45-7.69 (m,10H); 6.90 (d,	J=8Hz,2H); 6.63 (d,J= 8Hz,	1H); 5.17(d,J=6Hz,1H); 4.15-	4.0 (m,1H); 3.82 (s,2H); 3.5	(dd,J=6,14Hz,1H); 2.96(dd,	J=9,14Hz.1H)	mp=111-112ºC	MS CI+ (M+1)=426		mp=122-124ºC	mp=105-106°C	
C <sub>6</sub> H <sub>5</sub> -									2	>		4-(C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub> O)-	C <sub>6</sub> H <sub>4</sub> -
cis								trans	trans		trans	trans	
4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -								4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -		4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-(C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub> O)-C <sub>6</sub> H <sub>4</sub> -	
I								I	Ŧ		I	H	
S								C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -		C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	
7AQ								7AR	7AS		7AT	7AU	

				7		 ·	<del>-</del>		_
	o C	mp=186-187°C		mp=120-123°C	mp=243-245.5°C	mp=97-98,5°C	mp=78-81oC	mp=60-63°C	mp=108-110°C
	Box	C <sub>6</sub> H <sub>5</sub> -		C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
z z z z z z z z z z z z z z z z z z z	Rel. Stereo-	cis		cis	cls	cis		trans	cis
P. P	H <sub>23</sub> -G-	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -		4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	-С(О)ОСН2СН3	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -
	R22-F.	Ι		I	Ŧ	Ξ	C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -	C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub> -
	R <sub>21</sub> -E-	CH <sub>3</sub> O-C(O)- CH=C(CH <sub>3</sub> )-	-HZ	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>2</sub> - N(CH <sub>3</sub> )-		C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -	cyclo-C <sub>5</sub> H <sub>9</sub> -	C <sub>6</sub> H <sub>5</sub> -
	Ex.	2		¥	7	K	7AC	7AD	7AF

					<del>-</del>	_				
	elemental analysis: calc'd	C=76.56, H=8.80, N=3.31;	found C=76.37, H=8.90,	N=3.46	mn-114-1170C	0-711-411-0111	elemental analysis: calc'd	C=76.22, H=5.45, N=3.29;	found C=76.00, H=5.46,	NI 2 20
	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -				4-CH <sub>2</sub> O-C <sub>6</sub> H <sub>4</sub> - mp-114-1170C	700	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -			
	trans				cis		cis			
	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -				4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -		4-CH3O-C6H4-			
	I				C <sub>6</sub> H <sub>5</sub> -		<b>E</b>			
:	C10H21-	-			C <sub>6</sub> H <sub>5</sub> -CH <sub>2</sub> -	6		<b>&gt;</b> >		
700	) AG				/AH	7.01	1			

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#### **EXAMPLE 8**

Stir a mixture of 981mg (3.01mmol) of the cis  $\beta$ -lactam of Example 1AK, above, and 20mg 10% palladium on carbon in 10 mL ethyl acetate under 1atm hydrogen gas at room temperature for 24h. Filter the mixture through Celite and concentrate in vacuo. Chromatograph the residue over 50g SiO<sub>2</sub>, eluting with 40% ethyl acetate-hexane to obtain 818mg (92%) of the title compound as a yellowish solid, mp = 142.0 - 142.5°C.

Using a similar procedure, the following compounds shown in Tables 8 and 8A can be prepared, and using conventional procedures, the amino group of compound 8A and the amino group of compound 8B can be protected with a t-butoxycarbonyl (BOC) group to obtain compounds 8G and 8H, respectively:

	_							_			
		Data		4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=138 5-139 5oC	4-CH <sub>2</sub> O-C <sub>6</sub> H <sub>4</sub> - FAB MS (M±1)-387 1	mn=46-480C	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - m <sub>D</sub> =129 0-131 00 C		4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - mp=171 0-171 50C	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> - CI MS:(M+1)=487	•
		P4		4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	3-NH2-CeH4-	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -		4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH3O-C <sub>6</sub> H <sub>4</sub> -	
	Relative	Stereo-	chemistry	cis	trans	trans	trans		trans	cis	-
O N N N N N N N N N N N N N N N N N N N		∢		4-NH <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> -	4-NH <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -		4-OH-C <sub>6</sub> H <sub>4</sub> -	4((CH <sub>3</sub> ) <sub>3</sub> C-O-C(O)-	NH)-C <sub>6</sub> H <sub>4</sub> -
		Œ		I	H	I	Н		H	I	
		۵		C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	4-NH <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> -	(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	C <sub>6</sub> H <sub>5</sub> -(CH <sub>2</sub> ) <sub>3</sub> -	
		Ë.		8B	8C	8D	8E		8F	표	

		400	mn-136 19700	CI MS-(M+1)-207	160=(INT):0M D
		G.	4-CH <sub>2</sub> O-C <sub>2</sub> H <sub>4-</sub>	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> -	ł
, ×, E	Rel. Stereo-	chemistry	3S.4R	3S,4R	
R1-0		R <sub>23</sub> -G-	4-NH2-C <sub>6</sub> H <sub>4</sub> -	4((CH3)3C-O-C(O)-	NH)-C <sub>6</sub> H <sub>4</sub> -
		R <sub>22</sub> -F-	I	I	
		R <sub>21</sub> -E-	CH <sup>3</sup> CH <sup>5</sup> -	CH <sub>3</sub> CH <sub>2</sub> -	
		EX.	8A	<b>8</b>	

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directly in the next step.

#### **EXAMPLE 9**

STEP 1: Combine 5-phenylvaleric acid (89.9 g, 0.504 mol) and thionyl chloride (89.3 mL, 1.225 mol), heat to 70°C and reflux for 1h.

Vacuum distill (50 - 100 mm Hg) the excess thionyl chloride and add 200 mL of dry toluene to the resultant residue. Vacuum distill again, then add 188 mL of dry THF to the crude 5-phenylvaleroyl chloride and use the resulting solution directly in the next step.

STEP 2: Combine 76 g (0.4289 mol) of R-(+)-4-benzyl-

oxazolidinone and 1.3 L of dry THF under dry nitrogen atmosphere. Cool the resulting solution to -78°C and add 278 mL of a 1.6 M solution of n-butyllithium in hexane over 30-40 minutes. Stir for an additional 30 minutes, then add the solution of Step 1 over a period of 45 min. Allow the mixture to warm to 0°C and stir for 1 h. Quench the reaction mixture by adding 673.6 mL of K<sub>2</sub>CO<sub>3</sub> (1 M aqueous solution) and stir for 1 h. Distill off the THF under vacuum at 30-35°C. Dilute the residue with 1 L of water and extract with 3x800 mL of CH<sub>2</sub>Cl<sub>2</sub>. Combine the organic extracts and wash with 800 mL of water, then with 800 mL of brine. Dry the organic extracts over MgSO<sub>4</sub>, filter, then concentrate *in vacuo* to an oil. Dissolve the oil in 200 mL of hexane, then distill off the hexane under vacuum. Repeat the hexane treatment two more times, then dissolve the oil in 1.7 mL of CH<sub>2</sub>Cl<sub>2</sub>. The resulting solution is used

STEP 3: Cool the solution of Step 2 to -5°C to 0°C under dry nitrogen atmosphere. Add 129.8 mL of di-n-butylboron triflate, maintaining the temperature of the reaction mixture at -6°C to -3°C. Following the addition, stir the mixture for 10 min., then add 97.12 mL of diisopropylethylamine, again maintaining the temperature at -6°C to

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-3°C. Following the addition, stir the mixture at 0°C for 30 min., then cool the mixture to -78°C and stir for 30 min.. Add 57.4 mL of p-anisaldehyde and stir the mixture at -78°C for 30 min., then at 0°C for 1 h. While maintaining the temperature at 0°C to 5°C, quench the mixture by adding 688.2 mL of a pH 7 buffer solution (68 g KH<sub>2</sub>PO<sub>4</sub>, 12 g NaOH and 800 mL of water), then add 473 mL of 30% H<sub>2</sub>O<sub>2</sub> and stir the resulting mixture at 0°C for 1 h. Extract the mixture with 3x600 mL hexane:ethyl actetate (1:1), combine the organic extracts and wash with 800 mL of saturated NaHCO<sub>3</sub> (aqueous), then with 800 mL of brine. Dry the organic extracts over NaSO<sub>4</sub>, filter, and evaporate to an oil. Crystallize the oil from hexane/ethyl acetate (1:1) to give 176 g of the product as a white solid.

STEP 4: Combine the product of Step 3 (170 g, 0.36 Mol), 1595 mL of THF and 400 mL of water, stir the mixture and cool to about 3°C. Add 226 mL (2.156 Mol) of 30% H<sub>2</sub>O<sub>2</sub> to the mixture over 15 min., then add a solution of LiOH (36.2 g, 0.862 Mol) in 400 mL of water over a period of 30 min. Stir the reaction mixture at 0°C to 5°C for 3 h. Add a solution of 272 g of sodium sulfite in 850 mL of water over 70 min., keeping the temperature under 27°C. Concentrate the solvent under vacuum and add 7 L of water. Extract with 4x1.7 L of toluene. Acidify the aqueous layers to pH 2.4 with 3 N HCl. Extract with one 2.6 L portion and two 1.7 L portions of ethyl acetate. Combine the ethyl acetate extracts, wash with brine, dry over NaSO<sub>4</sub>, filter, then evaporate to give the product as a white solid, 112 g.

STEP 5: Combine the product of Step 4 (19.47 g, 62 mmol), 400 mL of acetonitrile, 9.49 g (62 mmol) of 1-hydroxybenzotriazole (HOBT), 22.91 g (186 mmol) of p-anisidine and 14.05 g (68.2 mmol) of dicyclohexylcarbodiimide (DCC). Stir the reaction mixture at 40°C for 4 h and confirm the consumption of starting material by TLC (6:4 hexane/ ethyl acetate). Concentrate the mixture to 1/3 its volume and partition between 300 mL of water and 300 mL of ethyl acetate. Filter the organic layer, then wash with 200 mL of 1 N HCl, then with two 100 mL portions of saturated NaHCO<sub>3</sub>, and two 100 mL portions of brine. Dry the

organic layer over NaSO<sub>4</sub> and concentrate to give the product as a brown solid, 24 g.

STEP 6: Combine the product of Step 5 (115 g, 0.2745 Mol) and 2.3 L of THF under dry nitrogen atmosphere and cool to -70°C. Stir the

5 mixture while adding a solution of 137 mL (0.551 Mol) of tri-n-butylphosphine in 113 mL THF, and 163 mL (1.03 Mol) of diethylazodicarboxylate (DEAD) over 2 h. Allow the mixture to warm to room temperature and stir overnight. Remove the solvent under vacuum. Filter the residue through a plug of silica gel using CH<sub>2</sub>Cl<sub>2</sub>/

10 hexane/ ethyl acetate (70:24:6) as the eluant. Evaporate the solvent and purify the residue by preparative HPLC (silica gel, 15% ethyl acetate/hexane) to give 88 g of the (3R,4S) enantiomerically pure title compound, [\alpha]<sub>D</sub><sup>25</sup> = -19.3° (MeOH).

## Example 10

The compound of Example 7AU (0.686 g, 0.0012 moles) dissolved in 1:1 ethanol:ethyl acetate is hydrogenated over 0.70 grams 10% Pd on carbon at 50 psi for 16 hours. The resulting product is chromatographed (silica gel, 80:20 hexane:ethyl acetate) to give 0.432 grams of the title compound, mp 160-161°C.

Using substantially the same procedure the following compound can be prepared:

Example 10A

Elemental Analysis
Calculated for C<sub>25</sub>H<sub>25</sub>NO<sub>3</sub>
C, 77.49; H, 6.50; N, 3.61
Found: C, 77.47; H, 6.46; N, 3.74

#### Example 11

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A flask is charged with the compound of Example 7J (27 g, 73.6 mmole), acetone (0.6 l), water (1.3 ml, 73 mmole), and p-toluenesulfonic acid-monohydrate (15.4 g, 81.1 mmole). The solution is stirred at 22°C for 5.5 h and the product collected by filtration (34.1 g). Recrystallization from methanol affords the title compound,26.0 g (79% yield), mp 200-202°C.

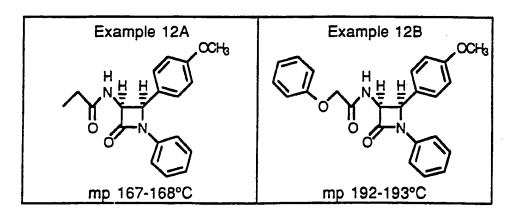
#### Example 12

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To a solution of the product of Example 11 (0.60 g, 1.36 mmole), triethylamine (0.14 g, 1.4 mmole) in CH<sub>2</sub>Cl<sub>2</sub> (2 ml) was added phenylacetylchloride (0.310 g, 2.0 mmole) dropwise over 0.5 h. After 5 h the reaction mixture was concentrated to dryness, and the residue purified by chromatography (silica, 2:1 hexane: ethyl acetate) to afford 0.27 g (51% yield) of the title compound, mp 217-218°C.

Using substantially the same procedure, the following compounds can be prepared:



## Example 13

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To a biphasic mixture of the product of Example 11 (0.80 g, 1.8 mmole), CH<sub>2</sub>Cl<sub>2</sub> (15 ml) and 30% aqueous tetrabutylammonium hydroxide (2.36 ml, 3.64 mmole) was added p-toluenesulfonylchloride (0.515 g, 2.7 mmole). After 2 h the reaction mixture was washed with 1X 1 N HCl, 2X H<sub>2</sub>O and the organic layer dried over MgSO<sub>4</sub>. Concentration followed by recrystillaztion from ethyl acetate afforded 0.38 g (50% yield) of the title compound, mp 204-205°C.

## Example 14

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To a solution of the product of Example 11 (0.8 g, 1.8 mmole), methanol (20 ml) and triethylamine (0.19 g, 1.8 mmole) was added phenylacetaldehyde (0.44 g, 3.6 mmole). After 0.25 h, NaBH<sub>3</sub>CN (0.172 g, 2.7 mmole) and ZnCl<sub>2</sub> (0.174 g, 1.3 mmole) were added to the reaction. After 3.5 h, the reaction was quenched with ammonium chloride, concentrated partially in vacuo, and dissolved in CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was washed 2X with water, dried and concentrated to afford crude product. Recrystallization from CH<sub>2</sub>Cl<sub>2</sub> afforded 0.44 g (65% yield) of the title compound, mp 138-139.5°C.

Using substantially the same procedure, the following compounds can be prepared:

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## Example 15

Add 49 mL (49 mmol) of 1M tetrabutylammonium fluoride 49 mL, 49 mmol, 1M in THF)to a room temperature solution of the compound of Example 1AS (5.1g, 9.8 mmol) in THF (25 mL). Allow the mixture to stir overnight. TLC (50% EtOAc/hexanes) indicates consumption of starting material. Transfer the reaction mixture to a separatory funnel, partion between saturated ammonium chloride and ether, extract with ether. Combine the etheral extracts, wash with water and brine, dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter and concentrate to a residue. Chromatograph (SIO<sub>2</sub>, 30-40% EtOAc/hexane) affords 3.31g (84%) of the title compound, mp 91-92°C.

The following compounds can be prepared via substantialy the same procedure:

Add MnO<sub>2</sub> (0.64g, 7.4 mmol) to a solution of the product of Example 15A (0.20g, 0.5 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (50 mL) at room temperature. Allow the mixture to stir overnight. TLC (20% EtOAc/hexanes) indicates consumption of starting material. Filter the mixture through celite washing the filtercake well with CH<sub>2</sub>Cl<sub>2</sub>. Concentrate the filtrate to obtain 0.19g (95%) of the title compound, mp 129-130 °C.

Using substantially the same procedure, the following compound can be prepared:

Example 16A

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#### Example 17

Add m-chloroperbenzoic acid (.16g, .092 mmol) to a solution of the compound of Example 1AU (0.31 g, 0.74 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (4 mL) at room temperature. Monitor the reaction by TLC (50% EtOAc/hexanes). Upon consumption of starting material (~1h), add Ca(OH)<sub>2</sub> (0.1g, 1.3 mmol) Stir the mixture for an additional 15 min..

Filter the mixture through celite washing the filtercake well with CH<sub>2</sub>Cl<sub>2</sub>. Concentrate the filtrate onto silica gel, using enough silica gel until a free flowing powder is obtained. Load the resulting powder onto a chromatography column loaded with silica gel and 5% MeOH/CH<sub>2</sub>Cl<sub>2</sub>. Elute with 5% MeOH/CH<sub>2</sub>Cl<sub>2</sub> to obtain 0.26 g of the title sulfoxide

(Example 17), mp 134-135°C and 0.50 g of the analogous sulfone: Example 17A

The following compounds can be prepared via substantially the same procedure:

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Reflux a mixture of the compound of Example 1AW (1.72 g, 3.49 mmol), 20 mL of THF and 3N HCl (20 mL) overnight. Monitor the reaction by TLC (50% EtOAc/hexanes). Upon consumption of starting material cool the mixture to room temperature, neutralize with saturated NaHCO<sub>3</sub> and extract with ethyl acetate. Combine the extracts, wash with water and brine, dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter and concentrate to a solid. Recrystallize from CH<sub>2</sub>Cl<sub>2</sub>/MeOH to obtain 0.322g of the title compound, mp 190.5-191.5 °C.

Using substantially the same procedure the following compound can be prepared:

Example 18A

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Example 19

Add a solution of NaBH<sub>3</sub>CN (0.19g, 3.0 mmol) and ZnCl<sub>2</sub>

(0.21g, 1.5 mmol) in methanol to a room temperature solution of the product of Example 16A (1.0g, 2.5 mmol) and morpholine (0.44 mL, 5.0 mmol) in THF. Monitor the reaction by TLC (50% EtOAc/hexanes).

Upon consumption of starting material the mixture was diluted with 0.1 N NaOH (120 mL) and remove the organic solvents on a rotary evaporator.

Extract the resulting solution with ethyl acetate. Combine the extracts, wash with water and brine, dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter and concentrate to a solid. Recrystallize from Et<sub>2</sub>O/hexanes to obtain 0.81g (69%) of the title compound, mp 100-101 °C.

The following compounds can be prepared by substantially

20 the same method:

Heat a solution of the compound of Example 1AR (1.10g, 2.95 mmol), 15% H<sub>2</sub>O<sub>2</sub> (0.45 mL) in acetic acid (3 mL) to a 100 °C for 3.5h with TLC monitoring (5% MeOH/CH<sub>2</sub>Cl<sub>2</sub>). Upon consumption of the starting compound, cool to room temperature, neutralize with sat. Na<sub>2</sub>CO<sub>3</sub> and dilute with ethyl acetate. Filter the mixture and wash the filtrate with water and brine, dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter and 10 concentrate to a residue. Chromatograph the residue (silica gel, 10% MeOH/CH<sub>2</sub>Cl<sub>2</sub>) to give the title compound, FAB MS (M+1) = 389.

### Example 21

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Add freshly prepared AgO (0.40 g, 1.73 mmol) to a solution of the product of Example 15A (0.46g, 1.15 mmol) and methyl iodide (0.21 mL, 3.45 mmol)in dry DMF (5 mL). Heat the mixture to 40-45 °C.with TLC monitoring (50% EtOAc/hexanes). Add proportional amounts of AgO and methyl iodide until TLC indicates consumption of starting compound. Cool to room temperature, particn between water and ethyl acetate, extract with ethyl acetate. Combine ethyl acetate

extracts, wash with water and brine, dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter and concentrate to a residue. Chromatograph the residue (silica gel, 30% EtOAc/hexanes) to give the title compound, FAB MS (M+1) = 416.

The following compound can be prepared by substantially

5 the same procedure:

Example 21A

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#### Example 22

Add freshly prepared Jone's reagen. (8 mL) dropwise to a solution of the product of Example 16 (2.76, 6.94 mmol) in acetone (80 mL) maintaining the temperature of the reaction mixture between 15-20 °C. Monitor by TLC (5% MeOH/CH<sub>2</sub>Cl<sub>2</sub>). Upon consumption of starting compound, quench the reaction with methanol. Concentrate to a residue and partion the residue between CH<sub>2</sub>Cl<sub>2</sub> and water. Extract with CH<sub>2</sub>Cl<sub>2</sub>, combine the extracts, wash with water, 10% Na<sub>2</sub>SO<sub>3</sub> (aq) and brine, then dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Filter and concentrate to afford 2.90g of the title compound, mp 64-65 °C.

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The following compound can be prepared by substantially the same procedure.

#### Example 22A

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# Example 23

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Add EDCI (0.2g, 1.03 mmoL) to a room temperature solution of the product of Example 22A (0.30, 0.72 mmol), N-methylmorpholine (0.11 mL, 0.94 mmol), morpholine (0.13 mL, 0.42mmol) and HOBT (0.12g, 0.87 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (8 mL). Stir the mixture overnight. Monitor by TLC (5% MeOH/CH<sub>2</sub>Cl<sub>2</sub>). Upon consumption of the starting compound, dilute with CH<sub>2</sub>Cl<sub>2</sub>, wash with 1M HCl, water, dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter and concentrate to a residue. Chromatogaph the residue (SiO<sub>2</sub>, 3% MeOH/CH<sub>2</sub>Cl<sub>2</sub>) to afford 0.3g (86% yield) of the title compound, mp 61-62 °C.

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Add EDCI (0.2g, 1.03 mmoL) to a room temperature solution of the product of Example 22A (0.30, 0.72 mmol), N-methylmorpholine (0.11 mL, 0.94 mmol), ethanol (0.1 mL, 1.44 mmol) and HOBT (0.12g, 0.87 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (8 mL). Stir the mixture overnight. Monitor by TLC (5% MeOH/CH<sub>2</sub>Cl<sub>2</sub>). Upon consumption of starting compound, dilute with CH<sub>2</sub>Cl<sub>2</sub>, wash with 1M HCl, water, dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter and concentrate to a residue. Chromatogaph the residue (SiO<sub>2</sub>, 3% MeOH/CH<sub>2</sub>Cl<sub>2</sub>) to afford 0.33g (100% yield) of the title compound, mp 76-77 °C.

The following compound can be prepared by substantially

# 15 the same procedure:

#### Example 24A

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# Example 25

Add methanesulfonyl chloride (0.05 mL, 0.67 mmoL) to a 0°C solution of the product of Example 8B (0.26, 0.67 mmol), pyridine(3 drops) in CH<sub>2</sub>Cl<sub>2</sub> (5 mL). Stir the mixture overnight. Monitor by TLC (50% EtOAc/hexanes). Upon consumption of starting material, dilute with CH<sub>2</sub>Cl<sub>2</sub>, wash with 0.5M HCl, 5% NaHCO<sub>3</sub>, water and brine, dry over anhydrous sodium sulfate, filter and concentrate to a residue. Chromatogaph the residue (SiO<sub>2</sub>, 50% EtOAc/hexanes) to afford 0.18g (58% yield) of the title compound, FAB MS (M+1) = 465.

By substantially the same procedure, the following

compounds can be prepared:

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Add acetyl chloride (0.04 mL, 0.57 mmoL) to a solution of the product of Example 8C (0.20, 0.52 mmol) and triethylamine(0.11 mL, 0.79) in CH<sub>2</sub>Cl<sub>2</sub> (3 mL) at room temperature . Monitor by TLC (50% EtOAc/hexanes). Upon consumption of starting material (~3h), dilute with CH<sub>2</sub>Cl<sub>2</sub>, wash with 5% NaHCO<sub>3</sub>, water and brine, dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter and concentrate to afford 0.19g, (85% yield) of the title compound, mp 153-154 °C.

Reflux a suspension of the product of Example 16 (0.6g, 1.5 mmol), hydroxylamine hydrochloride (0.4g, 4.5 mmol), sodium acetate (0.4g 4.5 mmol), methanol (12mL) and water (5mL) overnight. Monitor by TLC (50% EtOAc/hexanes). Upon consumption of starting material, evaporate to dryness, partition the residue between water and ethyl acetate, extract with EtOAc, combine the extracts, wash with water and brine, dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter and concentrate to a residue. Chromatograph the residue (silica gel, 20% EtOAc/hexanes) to provide the title compound, mp 98-99 °C; and Example 27A:

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Reflux a solution of the product of Example 17 (3.2g, 7.38 mmol), CH<sub>2</sub>Cl<sub>2</sub> (15 mL) trifluoracetic anhydride (15 mL) for 15 min. Monitor the reaction by TLC (100% EtOAc). Upon consumption of starting material, distill to remove most of solvent, cool to room temperature, and evaporate to a residue. Dissolve the residue in 50% triethylamine/methanol solution (30 mL) stir for 15 min., evaporate to dryness on a rotovap. Redissolve in CH<sub>2</sub>Cl<sub>2</sub>, wash with NH<sub>4</sub>Cl (sat.), dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter and concentrate onto enough silica gel (1g SiO<sub>2</sub>/ mmol substrate) that a free flowing powder results. Load the powder onto a chromatography column packed with silica gel and 30 % EtOAc/hexanes. Elute with 30-60% EtOAc/hexanes to provide 2.65g (89%) of a residue. Recrystallize from Et<sub>2</sub>O/CH<sub>2</sub>Cl<sub>2</sub> to give the title compound, mp 134-135°C.

#### Example 29

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Dissolve the product of Example 28 (1.81g, 4.49 mmol) in THF (25 mL) cool to 0 °C. Add 1N NaOH (5.39 mL, 5.39 mmol), and a

solution of aqueous ammonia (16.8 mL, 5.39 mmol), followed by cholorox (6.7 mL, 5.39 mmol). Monitor the reaction by TLC (50% EtOAc/hexane). Upon consumption of starting material, dilute with EtOAc, wash with water and brine, dry over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrate to provide the sulfinamine (1.88g).

Dissolve the sulfinamine (1.88g, 4.49 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (50 mL), add m-chloroperbenzoic acid (1.70g, 9.88 mmol). Monitor the reaction by TLC (50% EtOAc/hexane). Upon consumption of starting material, quench with solid Ca(OH)<sub>2</sub> (1.78g, 24.0 mmol), stir for 20 min., filter and concentrate onto enough silica gel (1g SiO<sub>2</sub>/ mmol substrate) that a free flowing powder results. Load the powder onto a chromatography column packed with silica gel and 50 % EtOAc/hexanes. Elute with 50% EtOAc/hexanes to provide 0.49g (24%) of the title compound, mp 174-176°C.

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#### Example 30

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Add BBr<sub>3</sub> (5.15 mL, 5.15 mmol, 1M in CH<sub>2</sub>Cl<sub>2</sub>) to a 0 °C solution of the compound of Example 7O (0.768g, 2.06 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (25 mL). Monitor the reaction by TLC (30% EtOAc). Upon consumption of starting material, quench with NaHCO<sub>3</sub> (sat.) and methanol, stir for 30 min., wash with NaHCO<sub>3</sub> (sat.) and water, dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter and concentrate onto enough silica gel (1g SiO<sub>2</sub>/ mmol substrate) such that a free flowing powder results. Load the powder onto a chromatography column packed with silica gel and 100% EtOAc. Elute with 100% EtOAc followed by 5% MeOH/EtOAc to provide 0.234g (32%) of the title compound, CI MS (M+1) = 359.

Beflux a mixture of the product of Example 16 (0.23g, 0.6 mmol) ethylene glycol (0.2mL, 3.6 mmol), p-toluene sulfonic acid and toluene (5 mL) overnight. Monitor by 1H NMR and TLC (50% EtOAc/hexanes). Cool to room temperature, wash with NaHCO<sub>3</sub>(sat.), water and brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and concentrate to give 0.27g (100%) of the title compound, EI MS (M) = 443.

The following compound can be prepared by substantially the same procedure:

Example 31A

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Stir a solution of 388mg (1.00 mmol) of the compound of Example 8F, and 180mg (1.20 mmol) chloramine-T Nal in 4.6 mL DMF. Add 377mg (1.2mmol) and stir for 18h at room temperature. Partition the reaction between 30 mL 1N HCl and 30 mL ethyl acetate. Wash the organic layer with 10% aqueous Na<sub>2</sub>SO<sub>3</sub> and dry the organic solution with MgSO<sub>4</sub>. Concentrate in vacuo and chromatograph the residue over (silica gel, 40% ethyl acetate-hexane) to give 170mg of the title compound, CI MS (M+1) = 514, and 41mg of the diodo analog (Example 32A):

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Stir 400 mg (0.96 mmol) of the compound of Example 8E, in 25 mL acetic acid and 5mL water at 0°C and add 133 mg (1.92 mmol) NaNO<sub>2</sub> in 4.7 mL water. Stir for 20 min. at 0°C add 212 mg (3.26 mmol) NaN<sub>3</sub> in 9 mL water and stir for 3h while warming to room temperature. Dilute with ethyl acetate and neutralize with NH<sub>4</sub>OH. Dry the ethyl acetate layer with MgSO<sub>4</sub> and concentrate to a residue. Recrystallize the residue from ether-hexanes to give the titled compound, mp = 80 - 84°C.

#### Example 34

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Dissolve 300 mg (0.77 mmol) of the compound of Example 8F, in 3 mL CH<sub>2</sub>Cl<sub>2</sub> and add 98 µL (120 mg, 0.80 mmol) 3carbomethoxypropionyl chloride and 146 µL (109 mg, 0.84 mmol)
Hünig's base. Stir at room temperature for 1.5h. Add another full portion of acid chloride and Hünig's base and stir for an additional hour.
Partition the reaction between ethyl acetate and 1N HCl. Dry the ethyl acetate layer with MgSO<sub>4</sub> and concentrate to a residue. Chromatograh

the residue ( $SiO_2$ , 40% ethyl acetate-hexane) and filter through grade I basic alumina eluting with 50%ethyl acetate-hexane to give 334 mg (86%) of the title compound, EI MS (M+) = 501.25.

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# Example 35

Benzyl bromide (0.44 g) was added to a solution of the compound of Example 15B (1.0 g) in acetone (15 mL) containing K<sub>2</sub>CO<sub>3</sub> (0.715 g). The reaction was heated at reflux for 26 h. The reaction mixture was then poured into water and the product extracted with ethyl acetate. The crude product was recrystallized from ethyl acetate/hexane to give (0.908 g, 74% yield) of the desired product, mp 115-116°C.

The following compounds were prepared via substantially the same procedure:

$$CIMS(M+1) = 474$$

Elemental analysis: calculated for C<sub>27</sub>H<sub>29</sub>NO<sub>3</sub>: C, 78.04; H, 7.03; N 3.37 Found: C, 78.00; H, 7.02; N, 3.55

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#### Example 36

The compound of Example 7AI (0.19 g) was cooled to -20°C. To this ws added a solution of diethyl zinc in toluene(4.3 mL of a 1.1 M solution) followed by diiodomethane (2.56 g). The reaction mixture was slowly allowed to warm to ambient temperature over three hours. Then the reaction mixture was warmed to 45°C for 5 min. After cooling the reaction mixture was treated with aqueous NH<sub>4</sub>Cl and product extracted with diethyl ether. The organic layer was washed with water, brine and concentrated to a residue. The residue was purified by chromatography (SIO<sub>2</sub>, ethyl acetate/hexane (3:7)) to give the title compound (0.17 g, 88% yield), Elemental analysis:

calculated - C, 78.42; H, 6.58; N, 3.39 found- C, 78.32; H, 6.48; N, 3.61.

### Example 37

step (a)

Combine 5-phenylvaleric acid (89.9 g, 0.504 mol) and SOCl<sub>2</sub> (89.3 mL, 1.225 mol) in a 500 mL round bottom flask equipped with a condenser and drying tube. Heat the flask to 70°C and maintain the reaction at reflux for 1h. Vacuum distill (50 - 100 mm Hg) the excess SOCl<sub>2</sub> and add 200 mL of dry toluene to the remaining mixture.

Vacuum distill a second time to remove the toluene and any residual SOCl<sub>2</sub>. Add 188 mL of dry THF to the crude acid chloride remaining in the reaction vessel and use the resulting solution directly in the next step.

step (b)

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Combine 76 g (0.4289 mol) of R-(+)-4-benzyl-oxazolidinone and 1.3 L of dry THF under dry nitrogen atmosphere. Cool the resulting solution to -78°C and add 278 mL of a 1.6 M solution of n-butyllithium in hexane over a period of 30-40 minutes. Stir the mixture for an additional 30 minutes following the addition. Add the solution of 5-phenylvaleroyl chloride from step (a) over a period of 45 minutes. Allow the mixture to warm to 0°C and stir for 1 h. Quench the reaction mixture by adding 673.6 mL of K<sub>2</sub>CO<sub>3</sub> (1 M aqueous solution) and stir for 1 h. Distill off the THF under vacuum at 30-35°C. Dilute the residue with 1 L of water and extract with three 800 mL portions of CH<sub>2</sub>Cl<sub>2</sub>. Combine the organic extracts and wash with 800 mL of water,

then with 800 mL of brine. Dry the organic extracts over MgSO<sub>4</sub>, filter, then concentrate *in vacuo* to an oil. Dissolve the oil in 200 mL of hexane, then distill off the hexane under vacuum. Repeat the hexane treatment two more times, then dissolve the oil in 1.7 mL of CH<sub>2</sub>Cl<sub>2</sub>. The resulting solution is used directly in the next step.

Cool the solution of the product from step (b) to -5°C to 0°C, under dry nitrogen atmosphere. Add 129.8 mL of di-n-butylboron 10 triflate at a rate that maintains the temperature of the reaction mixture at -6°C to -3°C. Following the addition, stir the mixture for 10 minutes, then add 97.12 mL of diisopropylethylamine at a rate that maintains the temperature of the reaction mixture at -6°C to -3°C. Following the addition, stir the mixture at 0°C for 30 minutes, then cool the mixture to 15 -78°C and stir for 30 minutes. Add 57.4 mL of p-anisaldehyde and stir the mixture at -78°C for 30 minutes, then at 0°C for 1 h. While maintaining the temperature at 0°C to 5°C, quench the mixture by adding 688.2 mL of a pH 7 buffer solution (68 g KH<sub>2</sub>PO<sub>4</sub>, 12 g NaOH and 800 mL of water), then add 473 mL of 30% H<sub>2</sub>O<sub>2</sub> and stir the 20 resulting mixture at 0°C for 1 h. Extract the mixture with three 600 mL portions of hexane:ethyl actetate (1:1). Combine the organic extracts and wash with 800 mL of saturated NaHCO3 (aqueous), then with 800 mL of brine. Dry the organic extracts over NaSO<sub>4</sub>, filter, and evaporate to an oil. Crystallize the oil from hexane/ethyl acetate (1:1) to give 176 g 25 of the product as a white solid.

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Combine the product of step (c) (170 g, 0.36 Mol), 1595 mL of THF and 400 mL of water, stir the mixture and cool to about 3°C. Add 226 mL (2.156 Mol) of 30% H<sub>2</sub>O<sub>2</sub> to the mixture over 15 minutes, then add a solution of LiOH (36.2 g, 0.862 Mol) in 400 mL of water over a period of 30 minutes. Stir the reaction mixture at 0°C to 5°C for 3 h. Add a solution of 272 g of Na<sub>2</sub>SO<sub>3</sub> in 850 mL of water over 70 minutes while keeping the temperature under 27°C. Distill off the bulk of the solvent under vacuum and add 7 L of water. Extract with four 1.7 L portions of toluene. Acidify the aqueous layers to pH 2.4 with 3 N HCl. Extract with one 2.6 L portion and two 1.7 L portions of ethyl acetate. Combine the ethyl acetate extracts, wash with brine, dry over NaSO<sub>4</sub>, filter, then evaporate to give the product as a white solid, 112 g.

Combine the product of step (d) (19.47 g, 62 mmol), 400 mL of acetonitrile, 9.49 g (62 mmol) of 1-hydroxybenzotriazole (HOBT), 22.91 g (186 mmol) of p-anisidine and 14.05 g (68.2 mmol) of dicyclohexylcarbodiimide (DCC). Stir the reaction mixture at 40°C for 4 h and confirm the consumption of starting material by TLC (6:4 hexane/ethyl acetate). Concentrate the mixture to 1/3 its volume and partition between 300 mL of water and 300 mL of ethyl acetate. Filter the organic layer, then wash with 200 mL of 1 N HCl, then with two 100 mL portions of saturated NaHCO<sub>3</sub>, and two 100 mL portions of brine. Dry the organic layer over NaSO<sub>4</sub> and concentrate to give the product as a solid, 24 g.

#### step (f)

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Combine the product of step (e) (115 g, 0.2745 Mol) and 2.3 L of THF under dry nitrogen atmosphere and cool to -70°C. Stir the mixture and simultaneously add a solution of 137 mL (0.551 Mol) of tri-n-butylphosphine in 113 mL THF, and 163 mL (1.03 Mol) of diethylazodicarboxylate (DEAD) over 2 h. Allow the mixture to warm to room temperature and stir overnight. Remove the solvent under vacuum. Filter the residue through a plug of silica gel using CH<sub>2</sub>Cl<sub>2</sub>/ hexane/ ethyl acetate (70:24:6) as the eluant. Evaporate the solvent and

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purify the residue by preparative HPLC (silica gel, 15% ethyl actetate/ hexane) to give 88 g (80% yield) of the  $\beta$ -lactam product.

Example 38

Cool a solution of 33.7 g (0.1 mol) of the product of Example 37, step (b), in 200 mL of CH<sub>2</sub>Cl<sub>2</sub> to -20°C. Stir the cold solution and add 11 mL (0.1 mol) of TiCl<sub>4</sub>. Stir the mixture for 10 min. at -20°C, then slowly add 30 mL (2 equiv.) of tetramethylethylenediamine (TMEDA) over a period of 10 min., while keeping the temperature below -10°C. Stir the mixture at -15° to -10°C for 70 min., then add 24 mL (2 equiv.) of p-anisaldehyde. Stir at -15° to -10°C for 1 hour, then allow the mixture to warm to 10°C while stirring for 40 min. Quench the reaction by adding 600 mL of 10% aqueous tartaric acid, then add 600 mL of ethyl acetate. Agitate well, then separate the layers, extracting the aqueous layer with another 200 mL of ethyl acetate. Combine the organic extracts and wash successively with water, saturated NaHCO<sub>3</sub> 20 (aqueous) and brine. Dry the organic solution over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter, then concentrate to a residue. Crystallize the residue from a mixture of 100 mL of ethyl acetate + 210 mL of hexane to give 36.8 g of the desired compound, which can be used in step (d) of Example 37.

# 5 Step (a)

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Dissolve 500 g (0.85 mol) of the product of Example 37, step(e), in 1700 mL of CH<sub>2</sub>Cl<sub>2</sub>, then add 4.0 g (12 mmol) of tera-n-butyl-ammonium hydrogen sulfate. Stir the mixture while cooling to 10° to 20°C and add 50% aqueous NaOH (200 g). Slowly add 60 g (285 mmol) of 2,6-dichlorobenzoyl chloride to the stirred mixture over a period of 30 min. Continue stirring at 15° to 20°C for 3 h., then pour the mixture into 2000 mL of cold water. Separate the layers and wash the organic layer with water until neutral pH is attained. Distill the

methylene chloride solution to reduce the volume to 800 mL. Heat the solution to reflux and add 800 mL of heptane. Cool the hot solution to 0°C to crystallize. Collect the product by filtration to give 116 g of the dichlorobenzoate product.

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Combine 500 g (0.85 mol) of the product of step (a) with 250 g (1.1 mol) of benzyltriethylammonium chloride, 2000 mL of CH<sub>2</sub>Cl<sub>2</sub> and 8000 mL of methyl t-butyl ether. Stir the mixture while cooling to 15° to 20°C, then add 1000 mL of 50% aqueous NaOH over a period of 10 min. Stir the mixture for 4 h., then pour into 5000 mL of water and 4 kg of ice. Separate the layers and wash the organic layer with water until the pH is neutral. Distill the solvent to reduce the volume to 2000 mL, then filter. Evaporate the filtrate to a residue and purify the residue by chromatography on silica gel to obtain the crude product. Crystallize the product from 6 volumes of a 1:2 mixture of methyl t-butyl ether and heptane at 0°C to give the product (240 g).

# 5 Step (a):

Dissolve 3.23 g (10 mmol) of the product of Example 37, step (b), in 50 mL of CH<sub>2</sub>Cl<sub>2</sub>, then stir under nitrogen atmosphere while cooling to -20°C. Add 10 mL (10 mmol) of a 1M solution of TiCl<sub>4</sub> in CH<sub>2</sub>Cl<sub>2</sub>, stir the mixture for 5 min., then add 1.5 mL (10 mmol) of TMEDA. Stir the mixture at -25° to -20°C for 1 h., then slowly add 4.8 g 5 (20 mmol) of the Schiff's base derived from anisaldehyde and panisidine as a solution in 50 mL of CH<sub>2</sub>Cl<sub>2</sub> over a period of 30 min. Stir the mixture at -20°C for 30 min, then gradually warm to 0°C. The reaction is monitored by HPLC (Zorbax® Sil column, 1:4 ethyl acetate/hexane), while stirring at 0°C, until complete. Quench the 10 mixture by pouring into 50 mL of 10% aqueous tartaric acid. Extract with ethyl acetate, then wash the organic extract successively with saturated NaHCO<sub>3</sub> (aqueous) and brine. Dry the organic solution over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filter, then concentrate to give the crude product. Crystallize from ethyl acetate/hexane to give the purified product. 15

A solution of 0.505 g (0.89 mmol) of the product of step (a)

in 25 mL of CH<sub>2</sub>Cl<sub>2</sub> is stirred at 0°C, then treated with 1.77 mL (1.77 mmol) of a 1M solution of sodium bistrimethylsilylamide in THF. Stir the mixture while warming to room temperature, then continue stirring until the starting material is gone as determined by HPLC (typically 1 to 1 1/2 h.) Quench the mixture into 10% tartaric acid (aqueous). Wash the organic layer successively with saturated NaHCO<sub>3</sub> (aqueous) and brine, then dry over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Filter and concentrate to give the title compound.

The following formulations exemplify some of the dosage forms of this invention. In each the term "active compound" designates a compound of formula I or II, preferably (3R,4S)-1,4-bis-(4-methoxyphenyI)-3-(3-phenylpropyI)-2-azetidinone. However, this compound may be replaced by an equally effective amount of other compounds of formula I or II.

# EXAMPLE A Tablets

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No.	Ingredient	mg/tablet	mg/tablet
1 2 3	Active Compound  Lactose USP  Corn Starch, Food Grade, as a 10%	100 122 30	500 113 40
4 5	paste in Purified Water Corn Starch, Food Grade Magnesium Stearate Total	45 <u>3</u> 300	40 Z 700

# Method of Manufacture

Mix Item Nos. 1 and 2 in suitable mixer for 10-15 minutes. Granulate the mixture with Item No. 3. Mill the damp granules through a coarse screen (e.g., 1/4", 0.63 cm) if necessary. Dry the damp granules. Screen the dried granules if necessary and mix with Item No. 4 and mix for 10-15 minutes. Add Item No. 5 and mix for 1-3 minutes. Compress the mixture to appropriate size and weight on a suitable tablet machine.

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# EXAMPLE B Capsules

No.	Ingredient	mg/tablet	mg/tablet
1	Active Compound	100	500
2	Lactose USP	106	123
3	Corn Starch, Food Grade	40	70
4	Magnesium Stearate NF	4	Z
	Total	250	700

#### 5 Method of Manufacture

Mix Item Nos. 1, 2 and 3 in a suitable blender for 10-15 minutes. Add Item No. 4 and mix for 1-3 minutes. Fill the mixture into suitable two-piece hard gelatin capsules on a suitable encapsulating machine.

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Using the test procedures described above, the following in vitro and in vivo data were obtained for the preferred compounds, which are referred to in the following table by the corresponding example numbers. For the in vitro ACAT data, negative percent inhibition denotes apparent stimulation, while positive numbers denote inhibition. For the in vivo results, data is reported as percent change versus control, therefore, negative numbers indicate a positive lipid-lowering effect.

ľ	ACAT (in vitro)		% Reduction (in vivo)			
Ex.	IC50	%	Conc.	Serum Cholest. Dose		
EX.	(mM)		(mM)	Cholest.	Esters	mg/kg
1		38	10	-45	-95	50
ı			, -	-17	-55	10
				0	0	5
1A	7.5	83	25	-10	-26	100
1B		28	10	0	0	50
1C		22	10	-6	-15	50
1D		39	10	0	0	50
1E		3	10			
1F		61	10	0	20	50
1G		21	10	-11	0	50
1H		57	10	-21	-51	50
11	4.5	81	10			
1J	3.0	86	10			
1K		-70	10	0	0	50
1L		20	10	-31	-75	50
				-22	-34	10
				-25	-30	5
1M	7.0	80	10	-11	-31	25
1P		39	10	-19	-54	50
18		-19	10	0	-32	85
1T		-11	10	0	0	50
1U	19	58	10	0	0	50
1V		64	10			
1W		9	10			
1X		9	10	-12	0	50
1Y		50	10			
1Z		-15	10	-15	-39	50
1AA	-	-36	10			
1AB		17	10			
1AC		40	10	<b>∥-16</b>	-33	50

	_	_			1	
1AD		-5	10			
1AE		-7	10			
1AF		0	10			
1AG		-3	10	-16	-29	50
1AH		4	10			
1AI		27	10			
1AJ		-17	10	0	0	50
1AK	•••	-33	10			
1AL		-43	10	0	0	50
1AM				-12	-29	50
1AN				-34	-85	50
1AO	•			-33	-78	50
1AP				-51	-95	50
1AQ				-20	-22	50
1AR		-41	10	-22	-25	50
1AS	1.0	83	10	0	-19	50
1AT	5.0	70	10	0	-28	50
1AU		44	10	0	-21	50
1AV		7	10	15	0	50
1AW		68	10	-14	-16	50
1AX				-16	-52	50
1AY				-28	-78	50
1AZ				-37	-91	50
1BA				0	-0	50
1BB		-16	10	0	0	50
1BC		-2	10	0	0	50
1BD		17	10	0	-25	50
1BE		30	10	-10	-21	50
1BF		-17	10	0	-23	50
1BG	•	3	10	12	0	50
1BH		50	10	0-	0	50
1BI		59	10	-17	-45	50
1BJ		56	10	0	0	50
1BK		55	10	0	0	50

		_			. 1	
1BL		42	10	-10	0	50
1BM		38	10	0	-21	50
1BN	7.0			-16	0	50
1BO	5.4	79	10	-48	-93	50
1BP		58	10	-9	0	50
1BQ		35	10	0	-17	50
1BR				-15	-70	50
1BS	- <u>-</u> -	  		0	-43	50
1BT				0	0	50
1BU				-39	<i>-</i> 95	50
1BV				-11	0	50
1BW			 :	-9	-29	50
1BX				0	0	50
1BY				-18	-72	50
1BZ				-9	0	50
1CA				-12	0	50
1CB		33	10	0	0	50
1CC		46	10	0	0	50
1CD	7.0	84	10	0	-23	50
1CE		15	10	-19	-17	50
1CF		53	10	-23	-47	50
1CG		8	10	-30	-61	50
1CH				-49	-95	50
				-41	-90	10
				-24	-80	3
1CI		26	10	-8	-23	50
1CJ				0	0	50
1CK		-12	10	0	-28	50
1CL		89	10	0	-18	50
1CM				0	0	50
1CN		-28	10	0	0	50
2	6.0	82	10	-15	-37	5
				-28	-69	10
2A	25			<b>∥</b> 0	0	50

· 3		56	10	0	0	50
зА		33	10	-12	-29	50
3B	6.0	84	10	-11	-16	50
3C		-18	10			
3D	12	42	10	0	0	50
3E		-24	10	0	0	50
3F	1.7	82	10	0	-31	50
3G	5.0	74	10	9	0	50
3H		66	10	-10	31	50
31		43	10	-9	0	50
3J		45	10	0	-16	50
3K		19	10	0	0	50
3L		53	10	0	0	25
3M		13	10	0	0	50
3N		62	10	0	0	50
30		-28	10	0	0	50
3Q		3	10	-6	0	50
3R		-8	10	0	0	50
3T		-9	10			
3U		46	10	0	0	50
3V	6.5	60	10-	0	0	50
3W		56	10	0	0	50
3X		·3	10			
3Y		-33	10	0	0	50
4		16	10			
5	18	33	10	-29	-77.5	50
				-20	-72	25
				-21	-60	10
5A		-70	10			
5C		21	10	-17	-24	50
5D				-38	-95	50
				-41	-94	30
				-24	-77	10
5E				-38	-91	50

			n		ا مم	150
5F	]			-44	-98	50
	1			-15	-59	30
				-13	-25	10
5G				-44	-93	50
5H				-57	-96	50
5I				-26	-70	50
5J				-38	-90	50
5K				-28	-49	50
5L				-50	-95	10
				-39	-84	3
				-54	-64	1
5M				-8	-36	10
J141				-12	0	3
				-20	-50	50
5N		-23	10	-17	-55	50
5P	_	53	10	-11	-33	50
5Q		-36	10	0	-19	50
5R		18	10	-20	-21	50
5S		28	10	-52	-98	50
5T				-15	-48	50
5U				-48	-86	50
5V				-37	-77	50
5W	<b></b>	-16	10	0	-19	50
5X	<b></b>	38	10	0	-39	50
5Y				-29	0	50
				-16	0	50
5Z 5AA				-53	-93	50
	H	30	10	-28	-72	50
5AB	5	59	10	14	0	50
6				-49	-95	40
7		1		-41	-89.	
				-30	-50	3
- A				-22	-54	50
7A				-13	-45	50

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7C				-51	-95	50
7D				-35	-74	50
7E				-52	-93	50
7F				0	-26	50
7J				-21	-27	50
7K				0	-32	50
7L				0	0	50
7M				-38	-94	50
7N				-11	0	50
70				-14	-20	50
7P				-40	-95	50
7Q				-16	0	50
7R				-35	-80	50
7S				-27	-82	50
71				-49	-93	50
7U				-29	-60	50
<b>7</b> V				0	0	50
7W				12	0	50
7X				-26	-78	50
7Y				-53	-94	50
		Ì		-27	-79	10
				-22	-55	3
7Z				-21	-39	50
7AA				-10	-42	50
7AC		31	10	0	0	50
7AD	<b>l</b>	-22	10	-12	-12	50
. 7AE		-95	10	-17	-10	50
7AF	<b> </b>	53	10	23	-15	50
7AG	<b> </b>			0	-15	50
7AH		-5	10	16	16	50
7AI				-63	-95	50
				-20	-73	10
				-11	-46	3
7AJ				О	-18	50

7AK				12	-28	50
7AL				-13	0	50
7AM				-24	-70	50
7AN				-12	-42	50
7AO				-41	-90	50
7AP				-24	-82	50
7AQ				-52	-91	50
7AR				-32	-88	50
7AS				0	-20	50
7AT				-10	-32	50
8		-12	10			
8A		-17	10	0	0	50
8B		-5	10	-9	-35	50
8C		63	10	0	0	50
8E				0	-32	50
8F		8	10	-16	-48	50
_				-20	-45	25
9	26	30	20	-43	-93	10
				-21.5	-66	5
				-25	-68	3
10A		-37	10	-19	-58	50
11				-8	0	50
12				-12	0	50
12A				-14	-39	50
12B				0	-26	50
12C				0	0	50
12D				0	0	50
13				-18	0	50
14				30	-87	50
14A				31	-78	50
14B				0	0	50
14C				0	0	50
15		-6	10	-21	-46	50
15A		2	10	0	0	50

15B		-55	10	0	-18	50
16		42	10	-25	-29	50
16A		55	10	-30	-53	50
17		25	10	-17	-52	50
17A		38	10	-11	-25	50
17B				0	-39	50
18				0	0	50
19		24	10	-20	-51	50
19A		64	10	-20	-34	50
19B		53	10	-26	-31	50
19C				-9	-46	50
19D				-15	-18	50
19E				-19	-30	50
20				12	0	50
21				-27	-56	50
21A				0	-25	50
22				-17	-33	50
22A				-16	-22	50
23				-22	-40	50
24				-17	0	50
24A				-21	-76	50
25				-8	-34	50
25A				-13	0	50
25B				0	0	50
25C				-8	-46	50
26				-4	-26	50
27			***	-12	-36	50
27A				-15	-42	50
28				-19	0 .	50
29				-22	-24	50
31				-19	-52	50
31A				0	0	50
32				0	0	50
32A				0	0 .	50

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	4	4	$\mathbf{a}$	
•	1	1	U	-

	_			f t	i i	
33				0	0	50
i				-39	-94	50
34		4.0	10	-15	-32	50
35		18	10	li	-84	50
35A				-34	ļ -	ł
35B		65	10	<b>∥-53</b>	-99	50
36				-53	-94	50
JU 1	I <b>I</b> i		•	-		

We claim:

#### 1. A compound having the structural formula

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wherein

A is -CH=CH-B;

-C≡C-B;

-(CH<sub>2</sub>)<sub>p</sub>-X-B, wherein p is 0, 1 or 2 and X is a bond, -NH-

10 or  $-S(O)_{0-2}$ -;

heteroaryl, benzofused heteroaryl, W-substituted heteroaryl or W-substituted benzofused heteroaryl, wherein heteroaryl is selected from the group consisting of pyrrolyl, pyridinyl, pyrimidinyl, pyrazinyl, triazinyl, imidazolyl, thiazolyl, pyrazolyl, thienyl, oxazolyl and furanyl, and for nitrogen-containing heteroaryls, the N-oxides thereof, and wherein W is 1-3 substituents on the ring carbon atoms selected from the group consisting of lower alkyl, hydroxy lower alkyl, lower alkoxy, alkoxyalkyl, alkoxyalkoxy, alkoxycarbonylalkoxy, (lower alkoxyimino)lower alkyl, lower alkanedioyl, lower alkyl lower alkanedioyl, allyloxy, -CF3, -OCF3, benzyl, R14-benzyl, benzyloxy, R14-benzyloxy, phenoxy, R14-phenoxy, dioxolanyl, NO2, -NR10R11, NR10R11(lower alkyl)-, NR10R11(lower alkoxy)-, OH, halogeno, -NHC(O)OR5, -NHC(O)R5, R6O2SNH-, (R6O2S)2N-, -S(O)2NH2, -S(O)0-2R10, tert-butyldimethylsilyloxymethyl, -C(O)R12 and

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heteroaryl ring nitrogen atoms, when present, are selected from the group consisting of lower alkyl, lower alkoxy, -C(O)OR<sub>5</sub>, -C(O)R<sub>5</sub>, OH, NR<sub>10</sub>R<sub>11</sub>(lower alkyl)-, NR<sub>10</sub>R<sub>11</sub>(lower alkoxy)-, -S(O)<sub>2</sub>NH<sub>2</sub> and 2-(trimethylsilyl)ethoxymethyl;

30 -C(O)-B; or

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-(CH<sub>2</sub>)<sub>k</sub>- N  $R_{13}$  , wherein k is 1 or 2;

D is B'- $(CH_2)_mC(O)$ -, wherein m is 1, 2, 3, 4 or 5;

B'- $(CH_2)_{q}$ -, wherein q is 2, 3, 4, 5 or 6;

B'-(CH<sub>2</sub>)<sub>e</sub>-Z-(CH<sub>2</sub>) $_{\Gamma}$ , wherein Z is -O-, -C(O)-, phenylene,

5 -NR<sub>8</sub>- or -S(O)<sub>0-2</sub>-, e is 0, 1, 2, 3, 4 or 5 and r is 1, 2, 3, 4 or 5, provided that the sum of e and r is 1, 2, 3, 4, 5 or 6;

B'-(C2-C6 alkenylene)-; B'-(C4-C6 alkadienylene)-;

B'- $(CH_2)_t$ -Z- $(C_2$ - $C_6$  alkenylene)-, wherein Z is as defined above, and wherein t is 0, 1, 2 or 3, provided that the sum of t and the number of carbon atoms in the alkenylene chain is 2, 3, 4, 5 or 6;

B'- $(CH_2)_f$ -V- $(CH_2)_g$ -, wherein V is  $C_3$ - $C_6$  cycloalkylene, f is 1, 2, 3, 4 or 5 and g is 0, 1, 2, 3, 4 or 5, provided that the sum of f and g is 1, 2, 3, 4, 5 or 6;

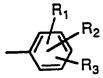
B'- $(CH_2)_t$ -V- $(C_2$ - $C_6$  alkenylene)- or B'- $(C_2$ - $C_6$  alkenylene)-V- $(CH_2)_t$ -, wherein V and t are as defined above, provided that the sum of t and the number of carbon atoms in the alkenylene chain is 2, 3, 4, 5 or 6;

 $B'-(CH_2)_a-Z-(CH_2)_b-V-(CH_2)_d-$ , wherein Z and V are as defined above and a ,b and d are independently 0, 1, 2, 3, 4, 5 or 6, provided that the sum of a, b and d is 0, 1, 2, 3, 4, 5 or 6;

 $$T\text{-}(CH_2)_{s^-}$, wherein T is cycloalkyl of 3-6 carbon atoms and s is 1, 2, 3, 4, 5 or 6; or$ 

naphthylmethyl, heteroarylmethyl, or W-substituted heteroarylmethyl, wherein heteroaryl and W are as defined above;

25 B is



B' is naphthyl, heteroaryl or W-substituted heteroaryl, wherein heteroaryl is as defined above, or

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R is hydrogen, fluoro,  $C_1$ - $C_{15}$  alkyl,  $C_1$ - $C_{15}$  alkenyl,  $C_1$ - $C_{15}$  alkynyl, or B-(CH<sub>2</sub>)<sub>h</sub>-, wherein h is 0, 1, 2, or 3;

R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> are independently selected from the group consisting of H, lower alkyl, hydroxy lower alkyl, lower alkoxy, alkoxyalkyl, alkoxyalkoxy, alkoxycarbonylalkoxy, (lower alkoxyimino)-lower alkyl, lower alkanedioyl, lower alkyl lower alkanedioyl, allyloxy, -CF<sub>3</sub>, -OCF<sub>3</sub>, benzyl, R<sub>14</sub>-benzyl, benzyloxy, R<sub>14</sub>-benzyloxy, phenoxy, R<sub>14</sub>-phenoxy, dioxolanyl, NO<sub>2</sub>, -NR<sub>10</sub>R<sub>11</sub>, NR<sub>10</sub>R<sub>11</sub>(lower alkyl)-, NR<sub>10</sub>R<sub>11</sub>(lower alkoxy)-, OH, o-halogeno, m-halogeno, -NHC(O)OR<sub>5</sub>, -NHC(O)R<sub>5</sub>, R<sub>6</sub>O<sub>2</sub>SNH-, (R<sub>6</sub>O<sub>2</sub>S)<sub>2</sub>N-, -S(O)<sub>2</sub>NH<sub>2</sub>, -S(O)<sub>0-2</sub>R<sub>10</sub>, tert-

butyldimethylsilyloxymethyl, -C(O)R<sub>12</sub>, -CH<sub>2</sub>-N R<sub>13</sub> and

 $R_{13}$ , or  $R_1$  is hydrogen and  $R_2$  and  $R_3$ , together with adjacent carbon atoms to which they are attached, form a dioxolanyl ring;

R<sub>1</sub>', R<sub>2</sub>' and R<sub>3</sub>' are independently selected from the group consisting of H, lower alkyl, hydroxy lower alkyl, lower alkoxy, alkoxyalkoxy, alkoxyalkoxy, (lower alkoxyimino)-lower alkyl, lower alkanedioyl, lower alkyl alkanedioyl, allyloxy, -CF<sub>3</sub>, -OCF<sub>3</sub>, benzyl, R<sub>14</sub>-benzyl, benzyloxy, R<sub>14</sub>-benzyloxy, phenoxy, R<sub>14</sub>-phenoxy, dioxolanyl, NO<sub>2</sub>, -NR<sub>10</sub>R<sub>11</sub>, NR<sub>10</sub>R<sub>11</sub>(lower alkyl)-, NR<sub>10</sub>R<sub>11</sub>(lower alkoxy)-, OH, halogeno, -NHC(O)OR<sub>5</sub>, -NHC(O)R<sub>5</sub>, R<sub>6</sub>O<sub>2</sub>SNH-, (R<sub>6</sub>O<sub>2</sub>S)<sub>2</sub>N-, -S(O)<sub>2</sub>NH<sub>2</sub>, -S(O)<sub>0-2</sub>R<sub>10</sub>, tert-butyldimethyl-

silyloxymethyl, -C(O)R<sub>12</sub>,  $R_{13}$  and  $R_{13}$  and  $R_{13}$ , or  $R_{14}$  is hydrogen and  $R_{24}$  and  $R_{34}$ , together with adjacent carbon atoms to which they are attached, form a dioxolanyl ring;

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R4 is , wherein n is 0, 1, 2 or 3, indanyl, benzoturanyl, benzodioxolyl, tetrahydronaphthyl, pyridyl, pyrazinyl,

pyrimidinyl or quinolyl;

R<sub>5</sub> is lower alkyl, phenyl, R<sub>14</sub>-phenyl, benzyl or R<sub>14</sub>-benzyl;

R<sub>6</sub> is OH, lower alkyl, phenyl, benzyl, R<sub>14</sub>-phenyl or R<sub>14</sub>-benzyl;

R<sub>7</sub> is lower alkyl, lower alkoxy, OH, halogeno, -NR<sub>10</sub>R<sub>11</sub>,

-NHC(O)OR<sub>5</sub>, -NHC(O)R<sub>5</sub>, NO<sub>2</sub>, -CN, -N<sub>3</sub>, -SH, -S(O)<sub>0-2</sub>-(lower alkyl),

-COOR<sub>9</sub>, -CONR<sub>10</sub>R<sub>11</sub>, -COR<sub>12</sub>, phenoxy, benzyloxy, -OCF<sub>3</sub>, or tert-butyldimethylsilyloxy, and where n is 2 or 3, the R<sub>7</sub> groups can be the same or different;

R<sub>8</sub> is H, lower alkyl, phenyl lower alkyl, or -C(O)R<sub>9</sub>;
R<sub>9</sub> is H, lower alkyl, phenyl or phenyl lower alkyl;
R<sub>10</sub> and R<sub>11</sub> are independently selected from H and lower alkyl;

15 R<sub>12</sub> is H, OH, alkoxy, phenoxy, benzyloxy, -NR<sub>10</sub>R<sub>11</sub>, lower alkyl, phenyl or R<sub>14</sub>-phenyl;

R<sub>13</sub> is -O-, -CH<sub>2</sub>-, -NH- or -N(lower alkyl)-; and

 $R_{14}$  is 1-3 groups independently selected from the group consisting of lower alkyl, lower alkoxy, -COOH, NO2, -NR10R11, OH or halogeno;

or a pharmaceutically acceptable salt thereof.

- A compound of claim 1 wherein R is hydrogen.
- 3. A compound of claim 1 or 2 wherein D is B'- $(CH_2)_{q}$ -, B'- $(CH_2)_{e}$ -Z- $(CH_2)_{r}$ -, B'- $(C_2$ - $C_6$  alkenylene)- or B'- $(CH_2)_{f}$ -V- $(CH_2)_{g}$ -, wherein B', Z, V, q, e, r, f, and g are as defined in claim 1.
- 4. A compound of claim 3 wherein D is B'-(CH<sub>2</sub>)<sub>q</sub>- wherein B' is phenyl and q is 3 or 4; B'-(CH<sub>2</sub>)<sub>e</sub>-Z-(CH<sub>2</sub>)<sub>r</sub>- wherein B' is p-fluorophenyl or p-methoxyphenyl, e is zero, Z is -O-, and r is 2; B'-(C<sub>2</sub>-

 $C_6$  alkenylene)- is 3-phenyl-1-propenyl; or B'-(CH<sub>2</sub>)<sub>f</sub>-V-(CH<sub>2</sub>)<sub>g</sub>- wherein B' is phenyl, f is 1, V is cyclopropylene, and g is zero.

- 5. A compound of claim 1, 2, 3 or 4 wherein A is -(CH<sub>2</sub>)<sub>p</sub>-X-B wherein wherein X, B and p are as defined in claim 1.
  - 6. A compound of claim 5 wherein p is zero and X is a bond.
- A compound of claim 6 wherein R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> are selected from the group consisting of H, OH, lower alkoxy, NO<sub>2</sub>,
   alkoxyalkoxy, m-halogeno, lower alkyl lower alkandioyl, NR<sub>10</sub>R<sub>11</sub>(lower alkoxy)-, allyloxy, phenoxy, alkoxycarbonylalkoxy and -C(O)R<sub>12</sub>.
  - 7. A compound of claim 1, 2, 3, 4, 5 or 6 wherein R<sub>4</sub> is phenyl, R<sub>7</sub>-substituted phenyl or indanyl.
- 8. A compound of claim 7 wherein R<sub>7</sub> is selected from the group consisting of lower alkyl, lower alkoxy, halogeno, -OCF<sub>3</sub>, lower alkylthio, -NR<sub>10</sub>R<sub>11</sub>, -CN, OH, and -COR<sub>12</sub>.
- 9. A pharmaceutical composition comprising a cholesterollowering effective amount of a compound having the structure

wherein

R<sub>20</sub> is phenyl, W-substituted phenyl, naphthyl, W-substituted naphthyl, benzodioxolyl, heteroaryl, W-substituted heteroaryl, benzofused heteroaryl and W-substituted benzofused heteroaryl, wherein heteroaryl is selected from the group consisting of pyrrolyl, pyridinyl, pyrimidinyl, pyrazinyl, triazinyl, imidazolyl, thiazolyl, pyrazolyl, thienyl, oxazolyl and furanyl, and for nitrogen-containing heteroaryls, the N-oxides thereof;

 $\mathsf{R}_{21},\,\mathsf{R}_{22}$  and  $\mathsf{R}_{23}$  are independently selected from H or

R<sub>20</sub>;

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W is 1 to 3 substituents independently selected from the group consisting of lower alkyl, hydroxy lower alkyl, lower alkoxy, alkoxyalkyl, alkoxyalkoxy, alkoxycarbonylalkoxy, (lower alkoxyimino)-lower alkyl, lower alkanedioyl, lower alkyl lower alkanedioyl, allyloxy, -CF<sub>3</sub>, -OCF<sub>3</sub>, benzyl, R<sub>14</sub>-benzyl, benzyloxy, R<sub>14</sub>-benzyloxy, phenoxy, R<sub>14</sub>-phenoxy, dioxolanyl, NO<sub>2</sub>, -NR<sub>10</sub>R<sub>11</sub>, NR<sub>10</sub>R<sub>11</sub>(lower alkyl)-, NR<sub>10</sub>R<sub>11</sub>(lower alkoxy)-, OH, halogeno, -NHC(O)OR<sub>5</sub>, -NHC(O)R<sub>5</sub>, R<sub>6</sub>O<sub>2</sub>SNH-, (R<sub>6</sub>O<sub>2</sub>S)<sub>2</sub>N-, -S(O)<sub>2</sub>NH<sub>2</sub>, -S(O)<sub>0-2</sub>R<sub>10</sub>, tert-butyldimethyl-

silyloxymethyl, -C(O)R<sub>12</sub>,  $-CH_2-N$   $R_{13}$  and -C-N  $R_{13}$ ;

E, F and G are independently a bond; C3-C6 cycloalkylene; C1-C10 alkylene; C1-C10 alkenylene; C1-C10 alkynylene; an alkylene, alkenylene or alkynylene chain as defined substituted by one or more substituents independently selected from the group consisting of phenyl, W-substituted phenyl, heteroaryl and Wsubstituted heteroaryl, wherein heteroaryl is as defined above; an alkylene, alkenylene or alkynylene chain as defined interrupted by one or more groups independently selected from the group consisting of -O-, -S-, -SO-, -SO<sub>2</sub>-, -NR<sub>8</sub>, -C(O)-, C<sub>3</sub>-C<sub>6</sub> cycloalkylene, phenylene, Wsubstituted phenylene, heteroarylene and W-substituted heteroarylene; or an interrupted alkylene, alkenylene or alkynylene chain as defined substituted by one or more substituents independently selected from the group consisting of phenyl, W-substituted phenyl, heteroaryl and Wsubstituted heteroaryl; or one of R21-E and R22-F is selected from the group consisting of halogeno, OH, lower alkoxy, -OC(O)R5, -NR<sub>10</sub>R<sub>11</sub>, -SH or -S(lower alkyl);

 $R_5$  is lower alkyl, phenyl,  $R_{14}$ -phenyl, benzyl or  $R_{14}$ -benzyl;  $R_6$  is OH, lower alkyl, phenyl, benzyl,  $R_{14}$ -phenyl or  $R_{14}$ -

30 benzyl;

 $R_8$  is H, lower alkyl, phenyl lower alkyl or -C(O) $R_9$ ;  $R_9$  is H, lower alkyl, phenyl or phenyl lower alkyl;

alkyl;

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R<sub>10</sub> and R<sub>11</sub> are independently selected from H and lower

R<sub>12</sub> is H, OH, alkoxy, phenoxy, benzyloxy,

-NR<sub>10</sub>R<sub>11</sub>, lower alkyl, phenyl or R<sub>14</sub>-phenyl;

R<sub>13</sub> is -O-, -CH<sub>2</sub>-, -NH- or -N(lower alkyl)-;

 $R_{14}$  is 1-3 groups independently selected from the group consisting of lower alkyl, lower alkoxy, -COOH, NO2, -NR10R11, OH or halogeno;

provided that when G is a bond,  $R_{23}$  is not H, and provided

that when R<sub>23</sub> is W-substituted phenyl, W is not p-halogeno; or a pharmaceutically acceptable salt thereof;

in a pharmaceutically acceptable carrier.

- 10. A compound of claim 1 selected from those prepared in
  15 examples 1, 1A-1M, 1P, 1AM-1AZ, 1BA-1BZ, 1CA-1Cl, 1CM-1CO, 3, 3A-3O, 5, 5B-5M, 5O-5Z, 5AA-5AB, 7, 7A-7I, 7M-7Z, 7AA, 7AB, 7AE, 7AI-7AK, 7AM-7AU, 8B-8F and 8H.
- 11. A compound of claim 1 or 10 selected from those listed in 20 Table 1.
  - 12. A compound of claim 1, 10 or 11 which is (3R,4S)-1,4-bis-(4-methoxyphenyl)-3-(3-phenylpropyl)-2-azetidinone.
- 25 21. A composition of claim 9 wherein the compound administered is selected from those prepared in examples 1-40..
  - The use of a compound of claim 1 for the manufacture of a medicament for lowering serum cholesterol.
  - 23. The use of a compound of claim 1 for the manufacture of a medicament for inhibiting the enzyme acyl CoA:cholesterol acyl transferase.

- A process for the preparation of a pharmaceutical 24. composition as claimed in claim 9 which comprises admixing a compound as defined in claim 9 with a pharmaceutically acceptable carrier.
- A method of lowering serum cholesterol levels in a 25. mammal in need of such treatment comprising administering an effective amount of a composition of claim 9.
- 10 A method of inhibiting the enzyme acyl CoA:cholesterol 26. acyl transferase by administering an effective amount of a composition of claim 9 to a mammal in need of such treatment.
- A process for producing a compound of claim 1, wherein R 15 27 is H, and D and A have trans relative stereochemistry, comprising cyclizing a hydroxyamide of the formula

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wherein D, A and R4 are as defined in claim 1, by treating

with: (i) a dialkylazodicarboxylate and a

- trialkylphosphine; or
- (ii) a di- or tri-chlorobenzoyl chloride, an aqueous base and a phase transfer catalyst, then treating the resulting dior tri-chlorobenzoate with an aqueous base and a phase transfer catalyst; or
  - (iii) a dialkylchlorophosphate, an aqueous base and a phase transfer catalyst; or
    - (iv) a di- or tri-chlorobenzoyl chloride and an alkali metal hydride.

28. A process for producing a compound of claim 1 having the formula

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wherein D, A and R<sup>4</sup> are as defined in claim 1, comprising:

(a) reacting a carboxylic acid of the formula DCH<sub>2</sub>-COOH, wherein D is as defined above, with a chlorinating agent;

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(b) deprotonating a chiral oxazolidinone of the formula

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wherein  $R^{18}$  and  $R^{19}$  are independently selected from the group consisting of: hydrogen,  $C_1$ - $C_6$  alkyl, phenyl, naphthyl, substituted phenyl, substituted naphthyl, and benzyl; with a strong base and treating the resulting anion with the product of step (a);

- (c) enolizing the product of step (b) with either:
  - (i) a dialkylboron triflate and a tertiary amine base;
  - (ii) TiCl<sub>4</sub> and tetramethylethylenediamine (TMEDA)

or a mixture of TMEDA and triethylamine;

then condensing with an aldehyde of the formula A-CHO, wherein A is as defined above;

(d) hydrolyzing the product of step (c) with a base and hydrogen peroxide;

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- (e) condensing the product of step (d) with an amine of the formula R<sup>4</sup>NH<sub>2</sub>, wherein R<sup>4</sup> is as defined above, by treating with a dehydrative coupling agent, optionally adding an activating agent; and
  - (f) cyclizing the product of step (e) by treating with:

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(i) a dialkylazodicarboxylate and a trialkylphosphine; or

- (ii) a di- or tri-chlorobenzoyl chloride, an aqueous base and a phase transfer catalyst, then treating the resulting di- or tri-chlorobenzoate with an aqueous base and a phase transfer catalyst; or
- (iii) a dialkylchlorophosphate, an aqueous base and a phase transfer catalyst; or
- (iv) a di- or tri-chlorobenzoyl chloride and an alkali metal hydride.
  - 29. A process for producing a compound of claim 1, wherein R is hydrogen, and D and A have trans relative stereochemistry, comprising:
    - (a) enolizing a compound of the formula

with TiCl<sub>4</sub> and tetramethylethylenediamine (TMEDA), then condensing with an imine of the formula A-CH=N-R<sup>4</sup>, wherein A and R<sup>4</sup> are as defined above, and wherein the group

represents an chiral auxiliary of the formula

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wherein  $R^{18}$  and  $R^{19}$  are independently selected from the group consisting of: hydrogen,  $C_1$ - $C_6$  alkyl, phenyl, naphthyl, substituted phenyl, substituted naphthyl, and benzyl; and

(b) cyclizing the product of step (a) by treating with a strong non-nucleophilic base.

30. A process for preparing compounds of claim 1 comprising:

(a) treating an ester of the formula D-CH(R)-C(O)OR<sup>17</sup>, wherein R<sup>17</sup> is lower alkyl, menthyl, or 10-(siisopropylsulfonamido)-isobornyl, with a strong base, and then with an imine of the formula A-CH=N-R<sup>4</sup>, wherein A and R<sup>4</sup> are as defined in claim 1; or

- (b) for preparing a compound of claim 1 wherein R is not hydrogen, reacting a compound of claim 1, wherein R is hydrogen, with a strong base and an alkylating agent or an acylating agent; or
- (c) for preparing a compound of claim 1, wherein R is hydrogen, and D and A have cis relative stereochemistry, reacting a compound of claim 1, wherein R is hydrogen, and D and A have trans relative stereochemistry, with a strong base at -80° to -40°C, and then with a proton source; or
- (d) for preparing a compound of claim 1, wherein R is hydrogen, and D and A have trans relative stereochemistry, reacting a compound of claim 1, wherein R is hydrogen, and D and A have cis relative stereochemistry, with an alkali metal t-butoxide, then quenching with acid; or
  - (e) reacting a carboxylic acid derivative of the formula D-CH(R)-C(O)-Z, wherein D and R are as defined in claim

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$$\begin{array}{c} O \\ II \\ -O-P-OC_6H_5 \\ I \end{array}, \text{ or } CH_3 \\ \text{ , with an imine of } \\ \text{1, and Z is Cl,} \\ \end{array}$$

the formula A-CH=N-R $^4$ , wherein A and R $^4$  are as defined in claim 1, in the presence of a tertiary amine base.

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International S	Searching Authority EUROPEA	N PATENT OFFI	CE	Signature of Authorize CHOULY J		

	International Application No	
	NTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET)	Relevant to Claim No.
ategory °	Citation of Document, with indication, where appropriate, of the relevant passages	
·,A	EP,A,O 481 671 (MERCK & CO., INC.) 22 April 1992 see claims	1-9, 22-30
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### INTERNATIONAL SEARCH REPORT

Box i	Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)
This inter	national search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:
	Claims Nos.:  because they relate to subject matter not required to be searched by this Authority, namely:  Remark: Although claims 25,26 are directed to a method of treatment of  (diagnostic method practised on) the human/animal body the search has been  carried out and based on the alleged effects of the compound/composition.
2.	Claims Nos.: because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
	Claims Nos.: because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).
Box II	Observations where unity of invention is lacking (Continuation of item 2 of first sheet)
This inte	rnational Searching Authority found multiple inventions in this international application, as follows:
1.	As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. 🗌	As all searchable claims could be searches without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3.	As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
4.	No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:
Remark	on Protest  The additional search fees were accompanied by the applicant's protest.  No protest accompanied the payment of additional search fees.

## ANNEX TO THE INTERNATIONAL SEARCH REPORT ON INTERNATIONAL PATENT APPLICATION NO. US 9

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This amore lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information. 15/10/92

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